



## Reaction Rates and Equilibrium Study Guide

### Topics Covered:

Factors affecting reaction rates  
Collision theory of reaction rates  
Reaction progress diagrams (activation energy and transition states)  
Rate laws and the method of initial rates  
Chemical equilibrium  
Equilibrium expressions and constants and their meaning  
Product-favored vs. Reactant-favored  
Shifting equilibrium and Le Chatelier's Principle  
Quantitative calculations of equilibrium systems using ICE tables

### Practice Questions: [Click here for an answer key](#)

- Describe how each of the following factors would affect the rate of "particle collisions" in your response.
  - Temperature decrease
  - Addition of iodide ions,  $I^-$ , as a catalyst
  - Increased reactant concentration
  - Crushing a solid into a powder (with larger surface area)
- The reaction progress diagram shows the energy changes that take place in the reaction of carbon monoxide gas and nitrogen dioxide gas.



# Chemistry Reaction Rates Study Guide

**Seager/Slabaugh**



## **Chemistry Reaction Rates Study Guide:**

*Chemical Principles Study Guide/Solutions Manual* John Krenos, Joseph Potenza, Carl Hoeger, 2007-01-18 Written for general chemistry courses Chemical Principles helps students develop chemical insight by showing the connection between chemical principles and their applications

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**Chemical Principles Student's Study Guide & Solutions Manual** John Krenos, Joseph Potenza, Loretta Jones, Lynn Koplitz, Thomas Spence, 2004-03-19 This combination manual is designed to help students avoid common mistakes and understand the material better The solutions manual section includes detailed answers and explanations to the odd numbered exercises in the text

*Student Study Guide/Solutions Manual for Essentials of General, Organic, and Biochemistry* Denise Guinn, Rebecca Brewer, Rachel C. Lum, 2009-09-15 The Student Study Guide and Solutions Manual provides students with a combined manual designed to help them avoid common mistakes and understand key concepts After a brief review of each section s critical ideas students are taken through stepped out worked examples try it yourself examples and chapter quizzes all structured to reinforce chapter objectives and build problem solving techniques The solutions manual includes detailed solutions to all odd numbered exercises in the text

**Study Guide for Chang's Chemistry** Kenneth W. Watkins, 1988

**Chemistry, Study Guide** Bernice G. Segal, Peter S. Shenkin, 1989-02-14 This Second Edition of the first year chemistry text known for its clarity of exposition and its large number of illustrative worked problems contains a more rigorous treatment of electrochemistry chemical equilibrium and thermochemistry Worked examples now number over 300 and exercises over 1460

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