

Chemistry 338 2nd Semester Final Exam Review Sheet

District Assessment

| Element | Symbol | Atomic # | Protons | Neutrons | Electrons | Mass # | Charge |
|----------|---------------------|----------|---------|----------|-----------|--------|--------|
| Tungsten | $^{184}_{74}W^{+2}$ | 74 | 74 | 110 | 72 | 184 | +2 |

1. Name the compound
 - Fe_2O_3 – iron (III) oxide
 - Cl_2O_8 – dichlorine octaoxide
2. Write the formula
 - Trisulfur Dinitride - S_3N_2
 - Zinc Hydroxide - $Zn(OH)_2$
 - Tin (IV) oxide - SnO_2
3. The pressure of a gas is 843 mm Hg at 24°C. What will the pressure be at standard temperature? Volume is constant.

$$\frac{P_1}{T_1} = \frac{P_2}{T_2}$$

$$\frac{843 \text{ mm Hg}}{297 \text{ K}} = \frac{P_2}{273 \text{ K}}$$

$$P_2 = 775 \text{ mm Hg}$$
4. The volume of a gas is 81.7 mL when the pressure is 119 kPa and the temperature is 73°C. What is the volume at STP?

$$\frac{P_1V_1}{T_1} = \frac{P_2V_2}{T_2}$$

$$\frac{(119 \text{ kPa})(81.7 \text{ mL})}{346 \text{ K}} = \frac{(101.3 \text{ kPa})(V_2)}{273 \text{ K}}$$

$$V_2 = 73.7 \text{ mL}$$
5. Between Be and Ba, which has a larger ionization energy? Larger atomic radius?

Larger ionization energy - Be
Larger atomic radius - Ba
6. Balance the equation & classify the type of reaction:
 $2Al + 3 CuSO_4 \rightarrow 1Al_2(SO_4)_3 + 3 Cu$ single replacement
7. Predict the products & classify the type of reaction:
 $H_3PO_4 + 3 NaOH \rightarrow 3 H_2O + Na_3PO_4$ double replacement
8. Calculate the Molar Mass: $(NH_4)_2CO_3$
 96 g/mole
9. Convert 2.75 moles of N_2O_5 to molecules

$$\frac{2.75 \text{ moles}}{1} \times \frac{6.02 \times 10^{23} \text{ molecules}}{1 \text{ mole}} = 1.66 \times 10^{24} \text{ molecules } N_2O_5$$

Solutions

Chemistry Spring Semester Final Exam Review Sheet

Ruskin Skidmore Freer

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web because data structures and algorithms are effectively patterns for solving problems you want to add as many of them as you can to your skill set by doing so you will find you solve more problems and use the right tools for the job in a more elegant way and you will learn a heap of them in this course

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