

Blue = Equation, concept

Yellow = Term, Cycle Step

Lecture 15 Expectations

1.) Net Equation for Glycolysis

- a.) $\text{Glucose} + 2\text{NAD}^+ + 2\text{ADP} + 2\text{P}_i \rightarrow 2 \text{pyruvate} + 2\text{NADH} + 2\text{H}^+ + 2\text{ATP} + 2\text{H}_2\text{O}$
- b.) Conversion of Glucose into pyruvate
- c.) Synthesis of ATP and NADH
- d.) Highly conserved pathways

2.) Know the substrate, product, and enzyme for each of the 10 steps of glycolysis

a.) Step 1:

- i.) Substrate: Glucose + ATP
- ii.) Product: Glucose 6-phosphate
- iii.) Enzyme: Hexokinase

b.) Step 2:

- i.) Substrate: Glucose 6-phosphate
- ii.) Product: Fructose 6-phosphate
- iii.) Enzyme: Phosphoglucomutase

c.) Step 3:

- i.) Substrate: Fructose 6-phosphate
- ii.) Product: Fructose 1,6-bisphosphate
- iii.) Enzyme: Phosphofructokinase-1 (PFK-1)

d.) Step 4:

- i.) Substrate: Fructose 1,6-bisphosphate
- ii.) Products: Dihydroxyacetone phosphate + Glyceraldehyde 3-phosphate
- iii.) Enzyme: Aldolase

e.) Step 5:

- i.) Substrate: Dihydroxyacetone phosphate
- ii.) Product: Glyceraldehyde 3-phosphate
- iii.) Enzyme: Triose phosphate isomerase

f.) Step 6:

- i.) Substrates: Glyceraldehyde 3-phosphate + inorganic phosphate

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Martin Silberberg, Libby Weberg

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Biochemistry James R. McKee, Trudy McKee, 2019-07-15 Biochemistry The Molecular Basis of Life is an intermediate one semester text written for students on degree pathways in Chemistry Biology and other Health and Life Sciences Designed for students who need a solid introduction to biochemistry but are not specializing in the subject the text focuses on essential biochemical principles that underpin the modern life sciences and offers the most balanced coverage of chemistry and biology of any text on the market The text equips students with a complete view of the living state emphasizes problem solving and applies biochemical principles to the fields of Health Agriculture Engineering and Forensics to show students the relevance of their learning McKee and McKee is respected for its balance of biology and chemistry consistently placing biochemical principles into the context of the physiology of the cell and biomedical applications

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College Chemistry Peter C. Scott, 1980

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