



CHEMISTRY

CHEAT SHEET



NAMING SIMPLE COMPOUNDS

NAMING SIMPLE COMPOUNDS CHEAT SHEET

Naming Compounds with Their Formulas
(Show the ions involved in the compound)



Directions for naming:

1. Write the cation (metal) name first.
2. Write the anion (non-metal) name second, changing the ending to -ide.
3. If the cation is a transition metal, include the oxidation state in Roman numerals in parentheses.
4. If the anion is a polyatomic ion, use its name.
5. If the compound is a molecular compound, use prefixes to indicate the number of atoms of each element.
6. If the compound is an acid, use the prefix H- followed by the name of the anion.

Binary Compounds

First Element	Anion
H	hydrogen
C	carbon
N	nitrogen
O	oxygen
F	fluorine
Cl	chlorine
S	sulfur
Br	bromine
I	iodine
P	phosphorus
As	arsenic
Sb	antimony
Bi	bismuth
Sn	tin
Pb	lead

Common Ions and Their Names

Formula	Name
Na ⁺	Sodium
K ⁺	Potassium
Ca ²⁺	Calcium
Mg ²⁺	Magnesium
Al ³⁺	Aluminum
Fe ²⁺	Iron (II)
Fe ³⁺	Iron (III)
Cu ²⁺	Copper (II)
Cu ⁺	Copper (I)
Zn ²⁺	Zinc
Ag ⁺	Silver
Li ⁺	Lithium
Rb ⁺	Rubidium
Cs ⁺	Cesium
Ba ²⁺	Barium
Strontium	Strontium
Yttrium	Yttrium
Scandium	Scandium
Vanadium	Vanadium
Chromium	Chromium
Manganese	Manganese
Cobalt	Cobalt
Nickel	Nickel
Cadmium	Cadmium
Mercury	Mercury
Thallium	Thallium
Lead	Lead
Bismuth	Bismuth
Antimony	Antimony
Arsenic	Arsenic
Phosphorus	Phosphorus
Sulfur	Sulfur
Selenium	Selenium
Tellurium	Tellurium
Polonium	Polonium
Astatine	Astatine
Francium	Francium
Radium	Radium
Actinium	Actinium
Thorium	Thorium
Protactinium	Protactinium
Uranium	Uranium
Neptunium	Neptunium
Plutonium	Plutonium
Americium	Americium
Curium	Curium
Berkelium	Berkelium
Californium	Californium
Einsteinium	Einsteinium
Fermium	Fermium
Mendelevium	Mendelevium
Nobelium	Nobelium
Lanthanum	Lanthanum
Cerium	Cerium
Praseodymium	Praseodymium
Neodymium	Neodymium
Europium	Europium
Gadolinium	Gadolinium
Terbium	Terbium
Dysprosium	Dysprosium
Hoium	Hoium
Erbium	Erbium
Thulium	Thulium
Ytterbium	Ytterbium
Lutetium	Lutetium
Hafnium	Hafnium
Tantalum	Tantalum
Tungsten	Tungsten
Rhenium	Rhenium
Osmium	Osmium
Iridium	Iridium
Rhodium	Rhodium
Palladium	Palladium
Silver	Silver
Copper	Copper
Gold	Gold
Platinum	Platinum
Pt	Pt
Au	Au
Ag	Ag
Cu	Cu
Fe	Fe
Co	Co
Ni	Ni
Zn	Zn
Al	Al
Mg	Mg
Ca	Ca
Na	Na
K	K
Li	Li
Rb	Rb
Cs	Cs
Ba	Ba
Strontium	Strontium
Yttrium	Yttrium
Scandium	Scandium
Vanadium	Vanadium
Chromium	Chromium
Manganese	Manganese
Cobalt	Cobalt
Nickel	Nickel
Cadmium	Cadmium
Mercury	Mercury
Thallium	Thallium
Lead	Lead
Bismuth	Bismuth
Antimony	Antimony
Arsenic	Arsenic
Phosphorus	Phosphorus
Sulfur	Sulfur
Selenium	Selenium
Tellurium	Tellurium
Polonium	Polonium
Astatine	Astatine
Francium	Francium
Radium	Radium
Actinium	Actinium
Thorium	Thorium
Protactinium	Protactinium
Uranium	Uranium
Neptunium	Neptunium
Plutonium	Plutonium
Americium	Americium
Curium	Curium
Berkelium	Berkelium
Californium	Californium
Einsteinium	Einsteinium
Fermium	Fermium
Mendelevium	Mendelevium
Nobelium	Nobelium
Lanthanum	Lanthanum
Cerium	Cerium
Praseodymium	Praseodymium
Neodymium	Neodymium
Europium	Europium
Gadolinium	Gadolinium
Terbium	Terbium
Dysprosium	Dysprosium
Hoium	Hoium
Erbium	Erbium
Thulium	Thulium
Ytterbium	Ytterbium
Lutetium	Lutetium
Hafnium	Hafnium
Tantalum	Tantalum
Tungsten	Tungsten
Rhenium	Rhenium
Osmium	Osmium
Iridium	Iridium
Rhodium	Rhodium
Palladium	Palladium
Silver	Silver
Copper	Copper
Gold	Gold
Platinum	Platinum
Pt	Pt
Au	Au
Ag	Ag
Cu	Cu
Fe	Fe
Co	Co
Ni	Ni
Zn	Zn
Al	Al
Mg	Mg
Ca	Ca
Na	Na
K	K
Li	Li
Rb	Rb
Cs	Cs
Ba	Ba
Strontium	Strontium
Yttrium	Yttrium
Scandium	Scandium
Vanadium	Vanadium
Chromium	Chromium
Manganese	Manganese
Cobalt	Cobalt
Nickel	Nickel
Cadmium	Cadmium
Mercury	Mercury
Thallium	Thallium
Lead	Lead
Bismuth	Bismuth
Antimony	Antimony
Arsenic	Arsenic
Phosphorus	Phosphorus
Sulfur	Sulfur
Selenium	Selenium
Tellurium	Tellurium
Polonium	Polonium
Astatine	Astatine
Francium	Francium
Radium	Radium
Actinium	Actinium
Thorium	Thorium
Protactinium	Protactinium
Uranium	Uranium
Neptunium	Neptunium
Plutonium	Plutonium
Americium	Americium
Curium	Curium
Berkelium	Berkelium
Californium	Californium
Einsteinium	Einsteinium
Fermium	Fermium
Mendelevium	Mendelevium
Nobelium	Nobelium
Lanthanum	Lanthanum
Cerium	Cerium
Praseodymium	Praseodymium
Neodymium	Neodymium
Europium	Europium
Gadolinium	Gadolinium
Terbium	Terbium
Dysprosium	Dysprosium
Hoium	Hoium
Erbium	Erbium
Thulium	Thulium
Ytterbium	Ytterbium
Lutetium	Lutetium
Hafnium	Hafnium
Tantalum	Tantalum
Tungsten	Tungsten
Rhenium	Rhenium
Osmium	Osmium
Iridium	Iridium
Rhodium	Rhodium
Palladium	Palladium
Silver	Silver
Copper	Copper
Gold	Gold
Platinum	Platinum

Naming Compounds with a Central and Peripheral



Sodium Chloride

Directions for naming:

1. Write the cation (metal) name first.
2. Write the anion (non-metal) name second, changing the ending to -ide.
3. If the cation is a transition metal, include the oxidation state in Roman numerals in parentheses.
4. If the anion is a polyatomic ion, use its name.
5. If the compound is a molecular compound, use prefixes to indicate the number of atoms of each element.
6. If the compound is an acid, use the prefix H- followed by the name of the anion.

Common Ions

Formula	Name	Formula	Name
Na ⁺	Sodium	Cl ⁻	Chloride
K ⁺	Potassium	Br ⁻	Bromide
Ca ²⁺	Calcium	I ⁻	Iodide
Mg ²⁺	Magnesium	S ²⁻	Sulfide
Al ³⁺	Aluminum	O ²⁻	Oxide
Fe ²⁺	Iron (II)	F ⁻	Fluoride
Fe ³⁺	Iron (III)	ClO ₄ ⁻	Perchlorate
Cu ²⁺	Copper (II)	ClO ₃ ⁻	Chlorate
Cu ⁺	Copper (I)	ClO ₂ ⁻	Chlorite
Zn ²⁺	Zinc	ClO ⁻	Hypochlorite
Ag ⁺	Silver	NO ₃ ⁻	Nitrate
Li ⁺	Lithium	NO ₂ ⁻	Nitrite
Rb ⁺	Rubidium	SO ₄ ²⁻	Sulfate
Cs ⁺	Cesium	SO ₃ ²⁻	Sulfite
Ba ²⁺	Barium	CO ₃ ²⁻	Carbonate
Strontium	Strontium	PO ₄ ³⁻	Phosphate
Yttrium	Yttrium	PO ₃ ³⁻	Phosphite
Scandium	Scandium	AsO ₄ ³⁻	Arsenate
Vanadium	Vanadium	AsO ₃ ³⁻	Arsenite
Chromium	Chromium	SbO ₄ ³⁻	Antimonate
Manganese	Manganese	SbO ₃ ³⁻	Antimonite
Cobalt	Cobalt	BiO ₄ ³⁻	Bismutate
Nickel	Nickel	BiO ₃ ³⁻	Bismutite
Cadmium	Cadmium	VO ₄ ³⁻	Vanadate
Mercury	Mercury	VO ₃ ³⁻	Vanadite
Thallium	Thallium	CrO ₄ ²⁻	Chromate
Lead	Lead	CrO ₃ ²⁻	Chromite
Bismuth	Bismuth	MnO ₄ ⁻	Manganate
Antimony	Antimony	MnO ₃ ⁻	Manganite
Arsenic	Arsenic	MoO ₄ ²⁻	Molybdate
Phosphorus	Phosphorus	MoO ₃ ²⁻	Molybdate
Sulfur	Sulfur	WO ₄ ²⁻	Tungstate
Selenium	Selenium	WO ₃ ²⁻	Tungstate
Tellurium	Tellurium	ReO ₄ ⁻	Perrhenate
Polonium	Polonium	ReO ₃ ⁻	Perrhenate
Astatine	Astatine	BrO ₄ ⁻	Perbromate
Francium	Francium	BrO ₃ ⁻	Perbromate
Radium	Radium	BrO ₂ ⁻	Perbromate
Actinium	Actinium	BrO ⁻	Perbromate
Thorium	Thorium	IO ₄ ⁻	Periodate
Protactinium	Protactinium	IO ₃ ⁻	Periodate
Uranium	Uranium	IO ₂ ⁻	Periodate
Neptunium	Neptunium	IO ⁻	Periodate
Plutonium	Plutonium	HO ₂ ⁻	Hydroperoxide
Americium	Americium	HO ₂ ⁻	Hydroperoxide
Curium	Curium	HO ₂ ⁻	Hydroperoxide
Berkelium	Berkelium	HO ₂ ⁻	Hydroperoxide
Californium	Californium	HO ₂ ⁻	Hydroperoxide
Einsteinium	Einsteinium	HO ₂ ⁻	Hydroperoxide
Fermium	Fermium	HO ₂ ⁻	Hydroperoxide
Mendelevium	Mendelevium	HO ₂ ⁻	Hydroperoxide
Nobelium	Nobelium	HO ₂ ⁻	Hydroperoxide
Lanthanum	Lanthanum	HO ₂ ⁻	Hydroperoxide
Cerium	Cerium	HO ₂ ⁻	Hydroperoxide
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Neodymium	Neodymium	HO ₂ ⁻	Hydroperoxide
Europium	Europium	HO ₂ ⁻	Hydroperoxide
Gadolinium	Gadolinium	HO ₂ ⁻	Hydroperoxide
Terbium	Terbium	HO ₂ ⁻	Hydroperoxide
Dysprosium	Dysprosium	HO ₂ ⁻	Hydroperoxide
Hoium	Hoium	HO ₂ ⁻	Hydroperoxide
Erbium	Erbium	HO ₂ ⁻	Hydroperoxide
Thulium	Thulium	HO ₂ ⁻	Hydroperoxide
Ytterbium	Ytterbium	HO ₂ ⁻	Hydroperoxide
Lutetium	Lutetium	HO ₂ ⁻	Hydroperoxide
Hafnium	Hafnium	HO ₂ ⁻	Hydroperoxide
Tantalum	Tantalum	HO ₂ ⁻	Hydroperoxide
Tungsten	Tungsten	HO ₂ ⁻	Hydroperoxide
Rhenium	Rhenium	HO ₂ ⁻	Hydroperoxide
Osmium	Osmium	HO ₂ ⁻	Hydroperoxide
Iridium	Iridium	HO ₂ ⁻	Hydroperoxide
Rhodium	Rhodium	HO ₂ ⁻	Hydroperoxide
Palladium	Palladium	HO ₂ ⁻	Hydroperoxide
Silver	Silver	HO ₂ ⁻	Hydroperoxide
Copper	Copper	HO ₂ ⁻	Hydroperoxide
Gold	Gold	HO ₂ ⁻	Hydroperoxide
Platinum	Platinum	HO ₂ ⁻	Hydroperoxide

Common Names for Specific Compounds

Formula	Common Name	Formula	Common Name
NaCl	Table salt	NaOH	Sodium hydroxide
KCl	Saline	KOH	Potassium hydroxide
CaCl ₂	Plaster of Paris	Ca(OH) ₂	Slaked lime
Na ₂ CO ₃	Soda ash	NaHCO ₃	Baking soda
NaHCO ₃	Baking soda	Na ₂ SO ₄	Sodium sulfate
Na ₂ SO ₄	Sodium sulfate	Na ₂ SiO ₃	Sodium silicate

Compound Names Cheat Sheet

David Baud



Compound Names Cheat Sheet:

Internet Cheat Sheet Ned Snell, 1999 Que's Cheat Sheet Series offers the busy beginning PC user just the right amount of information to get started with the Internet. Each chapter has a Basic Survival section that teaches the essentials of that chapter. The author then has highlighted important content for the reader in a different color to stand out. Finally, the author gives the reader additional notes, tips, and shortcuts in the handwritten margin notes. All combine to give the new PC user a quick, easy way to get connected to the Internet for using and exploring.

Chemistry Workbook For Dummies Peter J. Mikulecky, Chris Hren, 2014-11-26 Hundreds of practice problems to help you conquer chemistry. Are you confounded by chemistry? Subject by subject, problem by problem, *Chemistry Workbook For Dummies* lends a helping hand so you can make sense of this often intimidating subject. Packed with hundreds of practice problems that cover the gamut of everything you'll encounter in your introductory chemistry course, this hands-on guide will have you working your way through basic chemistry in no time. You can pick and choose the chapters and types of problems that challenge you the most or you can work from cover to cover. With plenty of practice problems on everything from matter and molecules to moles and measurements, *Chemistry Workbook For Dummies* has everything you need to score higher in chemistry. Practice on hundreds of beginning to advanced chemistry problems. Review key chemistry concepts. Get complete answer explanations for all problems. Focus on the exact topics of a typical introductory chemistry course. If you're a chemistry student who gets lost halfway through a problem or worse yet doesn't know where to begin, *Chemistry Workbook For Dummies* is packed with chemistry practice problems that will have you conquering chemistry in a flash.

Safety Standards, 1965 **Teach Yourself to Read Modern Medical Chinese** Bob Flaws, 1999 **Reach-touch Typewriting** Frances Moon Butts, 1926 **Merriam-Webster's Secretarial Handbook** Merriam-Webster, Inc., 1993 An authoritative reference source for today's office professional. Covers all aspects of office management. Includes an introduction to computers and a guide to business English. Provides abundant practical examples.

Chemical Technicians' Ready Reference Handbook Gershon J. Shugar, Jack T. Ballinger, 1996-06 The definitive reference of laboratory safety, analytic procedures, and instrumentation techniques for the modern chemical laboratory. Cited in BCL3, the new edition contains expanded chapters on gas chromatography (GC) and high performance liquid chromatography (HPLC) and physical properties and testing methods, with a new chapter on thermal analytic methods as well as on electrophoresis. Also includes up-to-date information on the role of chemical laboratory technicians and chemical process operators in industry and current data on laboratory safety, chemical waste disposal, government regulations, and ISO 9000. Explains in detail the day-to-day procedures, techniques, and formulas of today's chemical laboratory. The new edition, 2nd, was 1981, emphasizing the importance of safety, has been expanded to include additional information on material safety data sheets, chemical waste disposal, Right to Know regulations, and the National Fire Protection Association codes. Also new is material on such topics as gas chromatography, high performance liquid chromatography, infrared spectroscopy, atomic

absorption spectroscopy and computers in the laboratory *Webster's Home & Office Desk Reference* Merriam-Webster, Inc,1997 Quick comprehensive answers to the most frequently asked questions for conducting business Also includes guidelines for buying and using personal computers desktop publishing and suggestions for efficient set up and use of telephones and other electronic equipment *Introduction to Environmental Data Science* Jerry Davis,2023-03-13

Introduction to Environmental Data Science focuses on data science methods in the R language applied to environmental research with sections on exploratory data analysis in R including data abstraction transformation and visualization spatial data analysis in vector and raster models statistics and modelling ranging from exploratory to modelling considering confirmatory statistics and extending to machine learning models time series analysis focusing especially on carbon and micrometeorological flux and communication *Introduction to Environmental Data Science* is an ideal textbook to teach undergraduate to graduate level students in environmental science environmental studies geography earth science and biology but can also serve as a reference for environmental professionals working in consulting NGOs and government agencies at the local state federal and international levels Features Gives thorough consideration of the needs for environmental research in both spatial and temporal domains Features examples of applications involving field collected data ranging from individual observations to data logging Includes examples also of applications involving government and NGO sources ranging from satellite imagery to environmental data collected by regulators such as EPA Contains class tested exercises in all chapters other than case studies Solutions manual available for instructors All examples and exercises make use of a GitHub package for functions and especially data **Fundamental Filing Practice** Irene Place,Estelle L. Popham,Harry N. Fujita,1973 *Progressive Indexing and Filing* Remington Rand Business Service. Library Bureau Division,1928 **The Century Dictionary: The Century cyclopedia of names** ,1910 **Chronolog** ,1990 **Computer Handling and Dissemination of Data** National Research Council Canada,Royal Society of Canada,1987 **Principles of Indexing and Filing** Laura Hanes Cadwallader,Sarah Ada Rice,1971 **Modern Machine Shop** ,1932 *The Education Index* ,1980 *English Mechanic and Mirror of Science* ,1896 **Chemical Engineering Progress** ,1955 **British Journal of Pharmacology and Chemotherapy** ,1960

Unveiling the Magic of Words: A Review of "**Compound Names Cheat Sheet**"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Compound Names Cheat Sheet**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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time the total pressure $p_a \times \text{mole fraction}_a$
the total pressure $p_a \times \text{mole fraction}_a$ is often denoted by x_a and

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the multiplication of the mole fraction by 100 gives the mole percentage
mole fraction is a unitless and dimensionless expression
advantages of molar fraction
mole fraction is not dependent on the temperature

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web nov 6 2023 $x_a \text{ mol a mol a mol b and } x_b \text{ mol b mol a mol b}$
note in the given mixture the sum of all the mole fractions present is equal to one $x_a + x_b + \dots + x_n = 1$
when the mole fractions are multiplied by 100 they give the mole percentage
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molar mass of carbon tet CCl_4 12 4 355 154 g mol 1

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web notice that the mole fraction has no units on it and is written as a decimal value do not change it to percent note of caution you could see the term mole percent it is simply the mole fraction multiplied by 100 for example in the problem just below the mole fraction of cinnamic acid is 0.2885 its mole percent would be 28.85

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web since the mole fraction of the O_2 is 0 050 it contributes 0 050 or if you will 5 0 of the total pressure thus total pressure is 0 21 0 050 4 2 atm problem 9 a sample of 1 43 g of helium and an unweighed quantity of O_2 are mixed in a flask at room temperature

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