

CHEMISTRY CHEAT SHEET

NAMING SIMPLE COMPOUNDS

NAMING SIMPLE COMPOUNDS Cheat Sheet

Naming Compounds with Two Monatomic Ions

1. **Identify the cation and anion.**
2. **Find the cation's name.**
3. **Find the anion's name.**
4. **Drop the -ium from the cation's name.**
5. **Drop the -ide from the anion's name.**
6. **Put the cation's name first, followed by the anion's name.**
7. **End with the suffix -ide.**

Monatomic Ions

Element	Monatomic Ion
Hydrogen	Hydrogen
Fluorine	Fluoride
Chlorine	Chloride
Bromine	Bromide
Iodine	Iodide
Sodium	Sodium
Magnesium	Magnesium
Aluminum	Aluminum
Phosphorus	Phosphate
Sulfur	Sulfide
Chromium	Chromate
Manganese	Manganate
Iron	Iron
Lead	Lead

Monatomic Ions with Suffixes

Element	Monatomic Ion
Hydrogen	Hydrogen
Fluorine	Fluoride
Chlorine	Chloride
Bromine	Bromide
Iodine	Iodide
Sodium	Sodium
Magnesium	Magnesium
Aluminum	Aluminum
Phosphorus	Phosphate
Sulfur	Sulfide
Chromium	Chromate
Manganese	Manganate
Iron	Iron
Lead	Lead

Monatomic Ions with Suffixes (Continued)

Element	Monatomic Ion
Hydrogen	Hydrogen
Fluorine	Fluoride
Chlorine	Chloride
Bromine	Bromide
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Aluminum	Aluminum
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Sulfur	Sulfide
Chromium	Chromate
Manganese	Manganate
Iron	Iron
Lead	Lead

Common Monatomic Ions (Continued)

Element	Monatomic Ion
Hydrogen	Hydrogen
Fluorine	Fluoride
Chlorine	Chloride
Bromine	Bromide
Iodine	Iodide
Sodium	Sodium
Magnesium	Magnesium
Aluminum	Aluminum
Phosphorus	Phosphate
Sulfur	Sulfide
Chromium	Chromate
Manganese	Manganate
Iron	Iron
Lead	Lead

Compound Names Cheat Sheet

David Baud

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Unveiling the Magic of Words: A Review of "**Compound Names Cheat Sheet**"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Compound Names Cheat Sheet**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book's central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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we used mole fractions in gen chem 1 to describe gas mixtures where the partial pressure of a gas was its mole fraction time the total pressure $p = \sum x_i p_i$ mole fraction $x_i = \frac{n_i}{n_{\text{total}}}$ moles total right sum $n = \sum n_i$ all $x_i = \frac{n_i}{n}$ mole fraction of substance a is often denoted by x_a and

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web solution total mass of the answer 100 g mass of benzene 30 g therefore mass of carbon tet 100 - 30 g benzene 70 g molar mass $C_6H_6 = 6 \times 12 + 6 \times 1 = 78$ g mol 1 hence C_6H_6 s number of moles $70 / 78 = 0.903$ mol molar mass of carbon tet $CCl_4 = 12 \times 4 + 16 \times 4 = 154$ g mol 1

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web jan 30 2020 ans the molality of solution is $1 \text{ mol} / 1 \text{ kg} = 1 \text{ mol} / 1000 \text{ g} = 0.001 \text{ mol/g}$ or 1 m example 02 11 11 g of urea nh_2CONH_2 was dissolved in 100 g of water calculate the molarity and molality of the solution given n 14 h 1 c 12 o 16 given mass of solute urea 11 11 g mass of solvent water 100 g 0.1 kg to find molarity of solution solution

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web you may work out the mole fractions on your own problem 5 what is the mass of a sample of a 0 449 molal kbr that contains 2 92 kg of water 1 molality moles solute divided by kilograms solute 0 449 mol kg x 2 92 kg x 1 31108 mol of kbr 2 moles times molar mass equals grams 1 31108 mol 119 0023 g mol 156 g kbr

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web calculate the mass percent mole fraction molality and molarity of hexane c 6 h 14 d 0 660 g cm 3 when it is mixed with heptane c 7 h 12 d 0 684 g cm3 in a 35 0 55 0 ml ratio assume that the volumes are additive

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web notice that the mole fraction has no units on it and is written as a decimal value do not change it to percent note of caution you could see the term mole percent it is simply the mole fraction multiplied by 100 for example in the problem just below the mole fraction of cinnamic acid is 0 2885 its mole percent would be 28 85

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