



Crane Hand Signals

OSHA 1926.1423



USE MAIN HOIST



USE AUXILIARY HOIST
(use the whipline)



HOIST (load)



LOWER (load)



RAISE BOOM



**RAISE THE BOOM AND
LOWER THE LOAD**



LOWER BOOM



**LOWER THE BOOM
AND RAISE THE LOAD**



STOP



EMERGENCY STOP



SWING
(boom)



RETRACT BOOM



(one hand signal)



MOVE SLOWLY

A hand is placed in front of the hand that is giving the action signal.

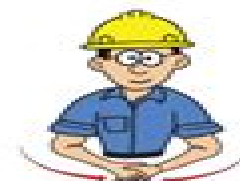
(Hoist slowly up is shown in this example.)



EXTEND BOOM



(one hand signal)



DOG EVERYTHING



**TRAVEL
ONE TRACK**



**TRAVEL
BOTH TRACKS**

Voice signals of crane must contain 3 elements, given in the following order:

- 1) Function (such as hoist, boom, etc.)
- 2) Direction, distance and/or speed
- 3) Function, stop command

example: "hoist, slow, slow, slow, hoist STOP"

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Crane Signals Posters

David Kirk



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 $\text{NaCl} + \text{F}_2 \rightarrow \text{NaF} + \text{Cl}_2$
 $2\text{H}_2\text{O} \rightarrow 2\text{H}_2 + \text{O}_2$
 $\text{Pb} + \text{HCl} \rightarrow \text{PbCl}_2 + \text{H}_2$
 $2\text{Al} + 3\text{Br}_2 \rightarrow 2\text{AlBr}_3$
 $\text{K}_2\text{SO}_4 + \text{Al}_2(\text{SO}_4)_3 \rightarrow \text{K}_2\text{O} + \text{Al}_2\text{O}_3 + \text{CO}_2$

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 $\text{GaBr} + \text{NaBr} \rightarrow \text{SNO}_4 + \text{NaSO}_3$
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 $3\text{HBr} + 1\text{KHCO}_3 + 1\text{H}_2\text{O} + 1\text{KBr} + \text{CO}_2 \rightarrow 2\text{GaBr}_3 + 3\text{Na}_2\text{SO}_3 + 1\text{Ga}_2\text{O}_3 + 3\text{SO}_3 + 6\text{NaBr} + 3\text{SNO}_2 + 3\text{NF}_3 + 3\text{NF}_2 + 1\text{N}_2\text{O}_3$ using the following equation
 $2\text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2\text{H}_2\text{O} + \text{Na}_2\text{SO}_4$

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 $\text{Ce} + \text{Zn} + \text{CuCl}_2 \rightarrow \text{ZnCl}_2 + \text{Cu} + \text{Zn}$
 how many moles of $\text{Ce} + \text{Zn} + \text{CuCl}_2$ will be produced from 23.0 g of $\text{Ce} + \text{Zn}$ assuming $\text{Ce} + \text{CuCl}_2$ is available in excess

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 $2\text{C}_6\text{H}_{10} + 17\text{O}_2 \rightarrow 12\text{CO}_2 + 10\text{H}_2\text{O}$
 $3\text{H}_2\text{O} + 1\text{KBr} + 1\text{CO}_2 \rightarrow 4\text{H}_2\text{O} + 2\text{GaBr}_3 + 3\text{Na}_2\text{SO}_3 + 1\text{Ga}_2\text{SO}_3 + 3\text{NaBr}$
 $5\text{S} + 3\text{NO}_2 + 3\text{NF}_3 \rightarrow 3\text{SnF}_2 + 1\text{N}_2\text{O}_3$
 using the following equation
 $\text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2\text{H}_2\text{O} + \text{Na}_2\text{SO}_4$

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 $3\text{H}_2\text{O} + 1\text{KBr} + 1\text{CO}_2 \rightarrow 4\text{H}_2\text{O} + 2\text{GaBr}_3 + 3\text{Na}_2\text{SO}_3 + 1\text{Ga}_2\text{SO}_3 + 3\text{NaBr}$
 solve the following stoichiometry grams grams problems 6

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