



- **ALL** questions are **MANDATORY**. You **MUST** solve.
- For molecular weight values refer to the attached periodic table.
- Conversion factors sheet is also attached to the exam.

Question 1 (35 Marks)

Choose the correct answer:

1. What basic SI unit is equal to $\text{kg} \cdot \text{m}^2/\text{s}^2$

- (a) Joule
- (b) Pascal
- (c) Tesla
- (d) Watt

2. One leg of a mercury U-tube manometer is connected to a pipe containing water under a gage pressure of 100 kPa. The mercury in this leg stands 750 mm below the water. What is the height of mercury in the other leg, which is open to the air? The specific gravity of mercury is 13.5

- (a) 0.2 m
- (b) 0.5 m
- (c) 0.8 m
- (d) 1 m

3. What are the products of complete combustion of a gaseous hydrocarbon?

- (a) Carbon monoxide only
- (b) Water, carbon monoxide, and carbon dioxide
- (c) Carbon dioxide and water
- (d) Carbon monoxide, water, and ammonia

4. Why is excess air required in combustion?

- (a) It allows the reaction to occur stoichiometrically
- (b) It reduces air pollution
- (c) It reduces the heat requirements
- (d) It allows complete combustion

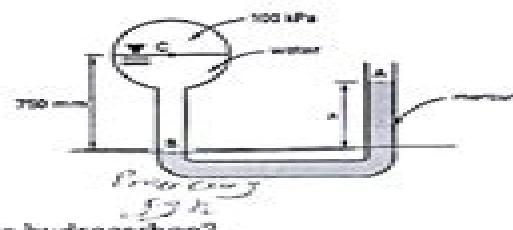
5. A sample of an unknown compound is found to be 49.5% carbon, 9.6% hydrogen, 19.2% nitrogen, and 21.9% oxygen by weight. What is its molecular formula?

- (a) $\text{C}_2\text{H}_4\text{NO}$
- (b) $\text{C}_2\text{H}_5\text{NO}$
- (c) $\text{C}_3\text{H}_5\text{N}_2\text{O}$
- (d) $\text{C}_3\text{H}_6\text{NO}$

6. Nitroglycerin is made by combining glycerol, nitric acid, and sulfuric acid. What are the minimum coefficients needed to balance the equation of this reaction?

$$\text{--- C}_3\text{H}_5\text{O}_3 + \text{--- HNO}_3 + \text{--- H}_2\text{SO}_4 \rightarrow \text{--- C}_3\text{H}_5\text{N}_3\text{O}_9 + \text{--- H}_2\text{O} + \text{--- H}_2\text{SO}_4$$

- (a) 2, 6, 2, 6, 2
- (b) 1, 3, 3, 1, 3, 2
- (c) 4, 2, 1, 1, 2, 4
- (d) 1, 3, 1, 1, 3, 1



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Michael R. Lindeburg

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