

Unit 1 Chemistry Study Guide

History of the Atom:

Democritus (450 BCE):

- Matter is not infinitely divisible.
- There are single particles that cannot be divided further, called **atomos**.

Aristotle (380 BCE):

- Matter is infinitely divisible.
- Matter is called **hyle**.

Antoine Lavoisier (1774):

- **Law of Conservation of Mass:** Mass cannot be created or destroyed.

Joseph Proust (1779):

- **Law of Definite Proportions:** All samples of a pure compound contain the same elements in the same proportion by mass (no matter quantity).

John Dalton (1803 – 1808):

- Observed work of Lavoisier and Proust
- Led Dalton to develop the first **atomic theory**.
- **Dalton's Atomic Theory:**
 - o All matter is composed of extremely small particles called atoms.
 - o All atoms of a given element are identical, having the same size, mass, and chemical properties. The atoms of one element are different from the atoms of all other elements.
 - o Atoms cannot be subdivided, created, or destroyed.
 - o Atoms of different elements can combine in simple whole number ratios to form chemical compounds.
 - o In chemical reactions, atoms are combined, separated or rearranged.
- **Law of Multiple Proportions:** Elements can combine in different proportions. However, the different compounds will have different properties but the ratio of atoms is always a whole number.

Billiard Ball Model:



J.J. Thompson (1906):

- **Cathode Ray Experiment –**
 - o Part 1: Thompson ran electricity through a cathode ray tube and when he put the negatively charged end of a magnet near the tube, he saw the ray of electricity

Chemistry 1 Study Guide

Mary Burnham, Carol Hurd



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Organic Chemistry Study Guide with Solutions Manual Neil E. Schore, 2007 The guide includes chapter introductions that highlight new material chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer

The United States Catalog Mary Burnham, Carol Hurd, 1928 *Pamphlet - Dept. of the Army* United States. Department of the Army, 1951 *All Hands*, 1968 **The American Educational Catalog**, 1920 **Fundamentals of Chemistry, Study Guide** James E. Brady, John R. Holum, 1988-04-20 This Third Edition of the widely used fundamentals textbook for science majors maintains the conversational writing style that made the previous editions so popular while including up to date treatments of important and current topics Emphasizes descriptive chemistry chemical reactions and properties while maintaining a solid treatment of chemical principles Common chemicals are used whenever possible as examples in both theoretical discussions and in problems and exercises Incorporates many pedagogical aids each chapter begins with a brief table of contents and each section begins with a preview of topics covered Chapters include frequent margin comments figures and photographs *Catalog of Copyright Entries. Third Series* Library of Congress. Copyright Office, 1978 *The United States Catalog*, 1906 **Student Study Guide and Solutions Manual to Accompany General, Organic, and Biochemistry** Katherine J. Denniston, Katherine, 2006 *The United States Catalog* Eleanor E. Hawkins, 1921 **The American Catalog**, 1885 **Laboratory Guide and Class Manual in Qualitative Chemical Analysis** Harry Sands Grindley, 1906 **Library of Congress Catalog: Motion Pictures and Filmstrips** Library of Congress, 1968 *The United States Catalog; Books in Print January 1, 1912* Marion Effie Potter, 1921

Organic Chemistry, Study Guide T. W. Graham Solomons, 1996 This supplement accompanies an updated text which features more than 350 new problems Carbonyl chemistry is now covered in consecutive chapters The concept of ionic reactions is consolidated before students move on to study radical reactions **Organic Chemistry Volume 1 Chemistry Study Guide** Bryant, 2019-09-13 **Study Guide to Organic Chemical Reactions** Milin Kurup, 2020-01-06 This Organic Study Guide and workbook was created by the author in hopes to help students have an organized collection of all reactions studied in Organic 1 and Organic 2 With multiple reactions in different categories an organized study guide can help students focus on certain reactions as needed Many students in this class also have hopes of using this study guide as an MCAT review guide this organized set can be useful for students to review before class exams and standardized test With comprehensive notes reactions and tips this study guide will help everyone succeed This was created by a renowned student Milin Kurup at the University of Florida double majoring in Microbiology and Cognitive and Behavioral Neuroscience As a student of Professor Dr Laura Peterson UF Chemistry Organic Division this guide included all notes a series of 140 chemical reactions and key mechanisms necessary for a comprehensive understanding of the subject **The Pharmaceutical Era**

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web oda ljubavi i razumevanju iz pera najpopularnije srpske autorke xx veka dubinom uvida u ljudske odnose njeni romani nalaze put do raznovrsnih čitalaca glavne ličnosti u ovom ljubavnom romanu su nežna plavokosa beograđanka ljljana i