

AP Chemistry Final Study Guide

Unit 1: Atomic Structure and Properties (7%–9%)

- At STP, one mole of gas always occupies 22.4 L
- **Mass Spectroscopy:**
 - Atoms have different isotopes (carbon-12, carbon-14, etc) and this will affect the average atomic mass
 - The most abundant isotope will have the greatest effect
- **Coulomb's Law:**
 - The greater the charge of the nucleus, the more energy is required to ionize an electron
 - The closer an electron is to the nucleus, the more energy is required to ionize an electron
- **Electromagnetic Radiation:**
 - When atoms absorb energy, electrons jump to higher energy levels
 - When electrons drop from higher to lower energy levels, atoms give off electromagnetic radiation energy
- **Kinetic Energy of Ionized Electrons:**
 - The faster an ejected electron is going, the more kinetic energy it has
 - Electrons originally further away from the nucleus require less energy to eject, so it will have more kinetic energy
- **Electron Configuration:**
 - When entering the d-block (transition metals), the energy level drops by 1 unless you are using shorthand notation
 - Nickel: $1s^2 2s^2 2p^6 3s^2 3p^6 3d^8 4s^2$ (notice the d-block)
 - Nickel (with shorthand): $[\text{Ar}] 4s^2 3d^8$
- **Law of Definite Proportions:**
 - A pure chemical compound broken down into elements always contains elements of a fixed ratio, independent of where and how it was created
- **Law of Multiple Proportions:**
 - When elements form compounds, the proportions of the elements in those chemical compounds can be expressed in small whole number ratios
- **Aufbau Principle:**
 - When building up the electron configuration of an atom, electrons are placed in orbitals, subshells, and shells in order of increasing energy

Chemistry Final Exam Study Guide

Xiaolong Qi



Chemistry Final Exam Study Guide:

AP Chemistry Exam Study Guide Accepted, 2019-09-19 **Study Guide for Chemistry** by Steven S. Zumdahl Martha B. Barrett, 1986 **Crimes Against the English Language**, **CliffsNotes AP Chemistry** Bobrow Test Preparation Services, 2009-02-09 The book itself contains chapter length subject reviews on every subject tested on the AP Chemistry exam as well as both sample multiple choice and free response questions at each chapter's end Two full length practice tests with detailed answer explanations are included in the book **Complete Book of Graduate Programs in the Arts and Sciences** Princeton Review (Firm), 2004-09 Our Best 357 Colleges is the best selling college guide on the market because it is the voice of the students Now we let graduate students speak for themselves too in these brand new guides for selecting the ideal business law medical or arts and humanities graduate school It includes detailed profiles rankings based on student surveys like those made popular by our Best 357 Colleges guide as well as student quotes about classes professors the social scene and more Plus we cover the ins and outs of admissions and financial aid Each guide also includes an index of all schools with the most pertinent facts such as contact information And we've topped it all off with our school says section where participating schools can talk back by providing their own profiles It's a whole new way to find the perfect match in a graduate school **NES Chemistry Study Guide** NES Chemistry Exam Prep Team, 2018-02-12 Imagine a study guide actually designed for teachers Because we know you've got a busy life we've developed a study guide that isn't like other certification materials out there With our NEW NES Chemistry Study Guide Test Prep and Practice Questions for the National Evaluation Series Chemistry Exam you get a swift but full review of everything tested on your certification exam FREE online resources are also included with your study guide Imagine having FREE practice questions digital flash cards study cheat sheets and 35 test tips available anytime anywhere on your cell phone or tablet Our resources will give you the push you need to pass your test the first time Because NES Chemistry Study Guide tells you everything you need to know and nothing you don't you'll have more time to concentrate on what's important in your life Furthermore NES Chemistry Study Guide gives you the opportunity to test your knowledge and assess your skills with practice tests and answers based on the actual exam Because teaching is more than a test we give you JUST what you need to succeed NES Chemistry Study Guide Test Prep and Practice Questions for the National Evaluation Series Chemistry Exam will provide you with a detailed overview of the NES Chemistry exam so you know exactly what to expect on test day We'll take you through all the concepts covered on the test and give you the opportunity to test your knowledge with practice questions Even if it's been a while since you last took a major test don't worry we'll make sure you're more than ready Cirrus Test Prep's NES Chemistry Study Guide Test Prep and Practice Questions for the National Evaluation Series Chemistry Exam includes a comprehensive REVIEW of Basic Principles of Matter Atomic and Nuclear Structure Bonding Naming Compounds Chemical Reactions Thermodynamics Solutions and Acid Base Chemistry Scientific Inquiry and Procedures as well as a FULL NES Chemistry

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