

## Sample Questions for the Chemistry Placement Test

The chemistry placement test is used to assess your present level of general chemistry knowledge in addition to your mathematical skills. You will be provided scratch paper (you cannot write on the test itself) and the Periodic Table of the Elements, and you will be allowed to use a calculator. You will have 45 minutes to answer 44 multiple choice questions, with the following sample questions intended only as a guideline:

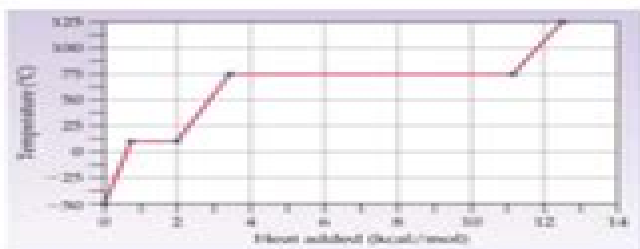
- If the formula for potassium chlorate is  $\text{KClO}_3$  and the formula for magnesium fluoride is  $\text{MgF}_2$ , then what is the formula for magnesium chlorate?
  - $\text{MgClO}_3$
  - $\text{Mg}_2\text{ClO}_3$
  - $\text{Mg}(\text{ClO}_3)_2$
  - $\text{Mg}_2(\text{ClO}_3)_6$
- From the periodic table, what is the atomic number of aluminum?
  - 26.98
  - 13
  - 18
  - 39.95
- Which one of the following elements does not exist as a diatomic molecule in nature?
  - hydrogen
  - nitrogen
  - fluorine
  - neon

For questions 4. and 5., consider the following reaction:



- The reaction can be classified as which one of the following types?
  - precipitation
  - decomposition
  - synthesis
  - double displacement
- How many moles of  $\text{Al}_2\text{O}_3$  can be produced from the reaction of 10.0 g of Al and 19.0 g of  $\text{O}_2$ ?
  - 0.581 mol
  - 0.371 mol
  - 0.185 mol
  - 0.396 mol
- What volume of 12.0 M HCl is required to make 75.0 mL of 3.50 M HCl?
  - 21.9 mL
  - 0.560 mL
  - 257 mL
  - 75.0 mL
- A fish tank holds  $1.029 \text{ yd}^3$  of water. What is this volume in cubic meters given that  $1 \text{ m} = 1.093 \text{ yd}$ ?
  - $1.062 \text{ m}^3$
  - $0.9414 \text{ m}^3$
  - $1.125 \text{ m}^3$
  - $0.7881 \text{ m}^3$
- Which one of the following is a strong acid?
  - $\text{HNO}_3$
  - $\text{CaSO}_4$
  - $\text{NH}_3$
  - $\text{NaOH}$

For questions 9. and 10., consider the following heating curve of a hypothetical substance:



- What is the boiling point of the substance?
  - $0^\circ\text{C}$
  - $-50^\circ\text{C}$
  - $12^\circ\text{C}$
  - $75^\circ\text{C}$
- What state of matter is the substance at  $50^\circ\text{C}$ ?
  - gas
  - liquid
  - solid
  - not enough information

### Answers

1. (c)      2. (b)      3. (d)      4. (c)      5. (c)      6. (a)      7. (d)      8. (a)      9. (d)      10. (b)

# Chemistry Placement Test Bucks Sample

**Annelies Wilder-Smith**



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