

Names & Formulas for Ionic Compounds

Give the name or formula of the following ionic compounds:

Name		Formula	
1)	Na_2CO_3 _____	21)	sodium phosphide _____
2)	NaOH _____	22)	magnesium nitrate _____
3)	MgBr_2 _____	23)	lead (II) sulfite _____
4)	KCl _____	24)	calcium phosphate _____
5)	FeCl_2 _____	25)	ammonium sulfate _____
6)	FeCl_3 _____	26)	silver cyanide _____
7)	$\text{Zn}(\text{OH})_2$ _____	27)	aluminum sulfide _____
8)	Be_2SO_4 _____	28)	beryllium chloride _____
9)	CrF_2 _____	29)	copper (I) arsenide _____
10)	Al_2S_3 _____	30)	iron (III) oxide _____
11)	PbO _____	31)	gallium nitride _____
12)	Li_3PO_4 _____	32)	iron (II) bromide _____
13)	TiI_4 _____	33)	vanadium (V) phosphate _____
14)	Co_3N_2 _____	34)	calcium oxide _____
15)	Mg_3P_2 _____	35)	magnesium acetate _____
16)	$\text{Ga}(\text{NO}_2)_3$ _____	36)	aluminum sulfate _____
17)	Ag_2SO_3 _____	37)	copper (I) carbonate _____
18)	NH_4OH _____	38)	barium oxide _____
19)	$\text{Al}(\text{CN})_3$ _____	39)	ammonium sulfite _____
20)	$\text{Be}(\text{CH}_3\text{COO})_2$ _____	40)	silver bromide _____

Compound Names And Formulas Activity Answer Key

Donada Peters, Edward I. Peters



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solving approach taken by the Chemical Principles text The new edition of the Study Guide includes many new worked out examples

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Study Guide with Selected Solutions Karen Timberlake, 2002-08-06 Written by the author the Study Guide is keyed to the learning goals in the text and designed to promote active learning through a variety of exercises with answers and mastery exams Also contains complete solutions to odd numbered problems

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