

Section Review 3.1

Part A Completion

| | |
|-----------------------|------------------------|
| 1. accuracy | 6. absolute value |
| 2. precision | 7. 100% |
| 3. experimental value | 8. scientific notation |
| 4. error | 9. known |
| 5. accepted value | 10. estimated |

Part B True-False

11. ST 13. AT
12. AT

Part C Matching

| | | |
|-------|-------|-------|
| 14. b | 16. a | 18. d |
| 15. c | 17. f | 19. e |

Part D Questions and Problems

20. a. 3
b. 4
21. a. 4.35×10^2 mL
b. 8.4 cm^2

c. 2
e. 2.1×10^2

Section Review 3.2

Part A Completion

| | |
|-----------|-------------------|
| 1. metric | 6. gram |
| 2. seven | 7. Celsius |
| 3. meter | 8. kelvin |
| 4. liter | 9. joule/calorie |
| 5. weight | 10. calorie/joule |

Part B True-False

11. NT 13. AT
12. NT 14. NT

Part C Matching

| | | |
|-------|-------|-------|
| 15. k | 20. l | 25. j |
| 16. e | 21. d | 26. f |
| 17. m | 22. c | 27. b |
| 18. g | 23. h | |
| 19. n | 24. i | |

Part D Questions and Problems

28. Volume = $1.8 \text{ cm} \times 8.8 \text{ cm} \times 30.5 \text{ cm}$
= $4.8 \times 10^3 \text{ cm}^3$

29. $^{\circ}\text{C} = \text{K} - 273 = 20 - 273 = -253^{\circ}\text{C}$

30. a. $\text{kg} \cdot 10^3$
b. $\text{kg} \cdot 10^3$

c. $\text{m} \cdot 10^{-3}$
d. $\text{m} \cdot 10^{-3}$

Section Review 3.3

Part A Completion

| | |
|----------------------|----------------------|
| 1. one (unity) | 6. steps |
| 2. conversion factor | 7. conversion factor |
| 3. remains the same | 8. denominator |
| 4. dimensional | 9. units |
| analysis | 10. unknown |
| 5. known | |

Part B True-False

11. NT 13. AT 15. AT
12. ST 14. NT

Part C Questions and Problems

16. a. $125 \text{ g} \times \frac{1 \text{ kg}}{1000 \text{ g}} = 0.125 \text{ kg}$
= $1.25 \times 10^{-1} \text{ kg}$

b. $0.12 \text{ L} \times \frac{1000 \text{ mL}}{1 \text{ L}} = 1.2 \times 10^2 \text{ mL}$

17. $\frac{1.0 \text{ } \mu\text{m}}{1500 \text{ cells}} \times \frac{2.54 \text{ } \mu\text{m}}{1 \text{ } \mu\text{m}} \times \frac{1 \text{ } \mu\text{m}}{100 \text{ } \mu\text{m}} \times \frac{10^6}{1 \text{ } \mu\text{m}}$
= $17 \text{ } \mu\text{m}/\text{cell}$

18. $186\,000 \text{ spores} \times \frac{1 \text{ s}}{1 \text{ s}} \times \frac{1.61 \text{ km}}{10^6 \text{ spores}} \times \frac{1 \text{ } \mu\text{m}}{1 \text{ spore}}$
= $2.99 \times 10^{-1} \text{ km}$

Section Review 3.4

Part A Completion

1. density 3. composition
2. intensive

Part B True-False

4. ST 5. ST

Part C Questions and Problems

6. $32.1 \text{ mL} \times \frac{1 \text{ cm}^3}{1 \text{ mL}} = 32.1 \text{ cm}^3$

7. Density = $\frac{\text{mass}}{\text{volume}} = \frac{127 \text{ g}}{32.1 \text{ cm}^3} = 3.96 \text{ g/cm}^3$

7. $1.00 \text{ L} \times \frac{1000 \text{ cm}^3}{1 \text{ L}} = 1000 \text{ cm}^3$

$1.58 \text{ kg} \times \frac{1000 \text{ g}}{1 \text{ kg}} = 1580 \text{ g}$

Density = $\frac{\text{mass}}{\text{volume}} = \frac{1580 \text{ g}}{1000 \text{ cm}^3} = 1.58 \text{ g/cm}^3$

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Joyit Mukherjee,Joyjit Mukherjee

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