



Chemical Reactions

Section 10.1 Reactions and Equations

For each statement, write an example of a chemical reaction or process. Write an equation for the reaction after chemical equation.

- _____ a. an element reacts with another element
- _____ b. an element reacts with a compound
- _____ c. a compound reacts with another compound
- _____ d. a compound reacts with an element
- _____ e. a compound reacts with a compound
- _____ f. a compound reacts with an element
- _____ g. a compound reacts with a compound
- _____ h. a compound reacts with an element
- _____ i. a compound reacts with a compound
- _____ j. a compound reacts with an element

Write the balanced chemical equation for each reaction. Write the names of the reactants, products, and any catalysts.

Reactants	Products	Conditions	Notes
_____	_____	_____	_____

Write the balanced chemical equation for the reaction of hydrogen gas with oxygen gas to form water. Write the names of the reactants, products, and any catalysts.

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Chemistry Study Guide For Content Mastery Key

**Michael L. Kamil, Peter B. Mosenthal, P.
David Pearson, Rebecca Barr**



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Chemistry, 2001-04-01 **Handbook of Reading Research, Volume III** Michael L. Kamil, Peter B. Mosenthal, P. David Pearson, Rebecca Barr, 2016-11-18 In Volume III as in Volumes I and II the classic topics of reading are included from vocabulary and comprehension to reading instruction in the classroom and in addition each contributor was asked to include a brief history that chronicles the legacies within each of the volume's many topics However on the whole Volume III is not about tradition Rather it explores the verges of reading research between the time Volume II was published in 1991 and the research conducted after this date The editors identified two broad themes as representing the myriad of verges that have emerged since Volumes I and II were published 1 broadening the definition of reading and 2 broadening the reading research program The particulars of these new themes and topics are addressed *Prentice Hall Chemistry*, 2000 **Organic**

Chemistry Francis A. Carey, 2000 A Market Leading Traditional Approach to Organic Chemistry Throughout all seven editions Organic Chemistry has been designed to meet the needs of the mainstream two semester undergraduate organic chemistry course This best selling text gives students a solid understanding of organic chemistry by stressing how fundamental reaction mechanisms function and reactions occur With the addition of handwritten solutions new cutting edge molecular illustrations updated spectroscopy coverage seamless integration of molecular modeling exercises and state of the art multimedia tools the 7th edition of Organic Chemistry clearly offers the most up to date approach to the study of organic chemistry Resources in Education, 2000-04 Saunders Interactive General Chemistry CD-ROM Mary L. Kotz, John C.

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