

# 1

# SOME BASIC CONCEPTS OF CHEMISTRY

Weightage 8 marks

1 mark - 1Q (Q No. 1)

2 mark - 1Q (Q No. 11)

5 mark - 1Q (Q No. 27)

## ONE MARK QUESTIONS

1. Following medicines are used in the treatment of which disease?

(a) Cisplatin (PUB)  
(b) Taxol (EQ)  
(c) Drug AZT (Azidothymidine) (EQ)

Ans: (a) Treatment of Cancer  
(b) Treatment of Cancer  
(c) Treatment of AIDS patients.

2. Give the S.I. unit of

(i) Density (PUB, May-17, Feb-17, 19, 20)  
(ii) Temperature (May 15)  
(iii) Amount of substance (Feb 18)  
(iv) Electric current (Feb 19)  
(v) Luminous intensity  
(vi) Volume (Feb 20)

Ans: (i) kg/m<sup>3</sup> (ii) Kelvin (iii) Mole  
(iv) Ampere (v) Candela (vi) m<sup>3</sup>

3. Express the following in scientific notation

(i) 0.00016 (Feb 18)

Ans:  $1.6 \times 10^{-4}$

(ii) 0.00085 (May 15)

Ans:  $8.5 \times 10^{-4}$

(iii) 0.00053 (Feb 16)

Ans:  $5.3 \times 10^{-4}$

(iv) 8008 (EQ)

Ans:  $8.008 \times 10^3$

(v) 0.001023 (PUB)

Ans:  $1.023 \times 10^{-3}$

(vi) 232.508 (Feb 19)

Ans:  $2.32508 \times 10^2$

4. How many significant figures are there in

(i) 0.00425 (Feb 17)  
(ii) 20.0  
(iii)  $6.022 \times 10^{23}$   
(iv) 0.2500 (PUB)  
(v) 3.10087  
(vi) 8.2015

Ans: (i) 3 (ii) 3 (iii) 4 (iv) 4 (v) 6 (vi) 5

5. Give the relation between

(a) Degree Fahrenheit and degree Celsius. (Feb 17)  
(b) Degree Celsius and Kelvin (EQ)

Ans: (a)  ${}^{\circ}\text{F} = 9/5 \times ({}^{\circ}\text{C}) + 32$   
(b)  $\text{K} = {}^{\circ}\text{C} + 273$

6. State law of Conservation of Mass (EQ)

Ans: It states that "matter can neither be created nor destroyed".

7. State law of Definite Proportions (PUB)

Ans: The law states that a given compound always contains exactly the same proportion of elements by weight.

8. State law of "Multiple Proportions". (EQ)

Ans: When two elements can combine to form more than one compound, the masses of one element that combine with a fixed mass of the other element, are in the ratio of small whole numbers.

9. State "Gay Lussac's Law" of gaseous volumes. (EQ)

Ans: The volume of gases taking part in a chemical reaction show simple whole number ratio to one another when all the volumes are measured at the same temperature and pressure.

10. State Avogadro's law. (May 14, Feb-17, 20)

Ans: It states that equal volumes of all gases contain equal number of molecules at same temperature and pressure.

# Chemistry Study Guide Questions And Concepts

**Robert C. Smoot, Smoot, Richard G.  
Smith, Jack Price**

## **Chemistry Study Guide Questions And Concepts:**

**Basic Concepts of Chemistry, Study Guide** Leo J. Malone, 1989-02-21 This Third Edition revised to provide smoother transitions between topics employs a concise yet informal approach to basic chemistry organized to help students employ basic math skills and problem solving strategies Writing style is straightforward and presentation incorporates many concrete analogies to clarify new concepts Includes many illustrative worked examples

**Robinson Chemistry Study Guide** Robinson, 1992

**Study Guide with Selected Solutions** Karen Timberlake, 2002-08-06 Written by the author the Study Guide is keyed to the learning goals in the text and designed to promote active learning through a variety of exercises with answers and mastery exams Also contains complete solutions to odd numbered problems

[Chemistry - Study Guide to Accompany Moore/Davies/Collins](#) Clare T. Furse, 1978

**Chemistry** Richard Post, Chad Snyder, Clifford C. Houk, 2020-09-16

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**A**

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organic chemistry clear and easily understood Each chapter ends with a list of frequently asked questions and answers followed by additional practice problems Answers are included in the Appendix *Study Guide for Organic Chemistry* Seyhan N. Ege, 1999 **NES Chemistry Study Guide** NES Chemistry Exam Prep Team, 2018-02-12 Imagine a study guide actually designed for teachers Because we know you've got a busy life we've developed a study guide that isn't like other certification materials out there With our NEW NES Chemistry Study Guide Test Prep and Practice Questions for the National Evaluation Series Chemistry Exam you get a swift but full review of everything tested on your certification exam FREE online resources are also included with your study guide Imagine having FREE practice questions digital flash cards study cheat sheets and 35 test tips available anytime anywhere on your cell phone or tablet Our resources will give you the push you need to pass your test the first time Because NES Chemistry Study Guide tells you everything you need to know and nothing you don't you'll have more time to concentrate on what's important in your life Furthermore NES Chemistry Study Guide gives you the opportunity to test your knowledge and assess your skills with practice tests and answers based on the actual exam Because teaching is more than a test we give you JUST what you need to succeed NES Chemistry Study Guide Test Prep and Practice Questions for the National Evaluation Series Chemistry Exam will provide you with a detailed overview of the NES Chemistry exam so you know exactly what to expect on test day We'll take you through all the concepts covered on the test and give you the opportunity to test your knowledge with practice questions Even if it's been a while since you last took a major test don't worry we'll make sure you're more than ready Cirrus Test Prep's NES Chemistry Study Guide Test Prep and Practice Questions for the National Evaluation Series Chemistry Exam includes a comprehensive REVIEW of Basic Principles of Matter Atomic and Nuclear Structure Bonding Naming Compounds Chemical Reactions Thermodynamics Solutions and Acid Base Chemistry Scientific Inquiry and Procedures as well as a FULL NES Chemistry practice test About Cirrus Test Prep Developed by experienced current and former educators Cirrus Test Prep's study materials help future educators gain the skills and knowledge needed to successfully pass their state level teacher certification exams and enter the classroom Each Cirrus Test Prep study guide includes a detailed summary of the test's format content and scoring an overview of the content knowledge required to pass the exam worked through sample questions with answers and explanations full length practice tests including answer explanations and unique test taking strategies with highlighted key concepts Cirrus Test Prep's study materials ensure that new educators feel prepared on test day and beyond **GACE Chemistry Study Guide** Trivium Test Trivium Test Prep, 2013-09-27 Think all GACE Chemistry study guides are the same Think again With easy to understand lessons and practice test questions designed to maximize your score you'll be ready You don't want to waste time and money retaking an exam You want to accelerate your education not miss opportunities for starting your future career Every year thousands of people think that they are ready for the GACE Chemistry exam but realize too late when they get their score back that they were not ready at all They weren't incapable

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Friedman,2017-07-09 This Study Guide was written specifically to assist students using the 6th edition of Introductory Chemistry It presents the major concepts theories and applications discussed in the text in a comprehensive and accessible manner for students It contains learning objectives chapter summaries and outlines as well as examples self tests and concept questions

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