

Intro to Chemistry Final Exam Study Guide

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- _____ 1. The Fahrenheit temperature scale is defined by which of the following temperatures?
- Water boils at 100 degrees and freezes at 32 degrees.
 - Water boils at 212 degrees and freezes at 32 degrees.
 - Water evaporates at 212 degrees and freezes at 0 degrees.
 - Liquid water turns to a gas at 100 degrees and to a solid at 0 degrees.
- _____ 2. A cold-blooded reptile basks on a warm rock in the sun. Its body is warmed by
- radiation.
 - conduction.
 - convection.
 - Both (a) and (b).
- _____ 3. Temperature is a measure of the average _____ energy of an object's particles.
- mechanical
 - kinetic
 - potential
 - light
- _____ 4. The temperature at which all molecular motion stops is
- 0°C.
 - 0°F.
 - 0 K.
 - All of the above
- _____ 5. Convert 500°F to degrees Celsius.
- 260°C
 - 296°C
 - 842°C
 - 958°C
- _____ 6. Convert 300 K to the Celsius scale.
- 0°C
 - 27°C
 - 73°C
 - 573°C
- _____ 7. Convert 458 K to the equivalent Fahrenheit temperature.
- 185°F
 - 237°F
 - 365°F
 - 731°F
- _____ 8. Which of the following temperatures is impossible to measure?
- 85°F
 - 50°C
 - 20 K
 - 545°F
- _____ 9. What is -175°C on the Kelvin scale?
- 76 K
 - 89 K
 - 98 K
 - 448 K
- _____ 10. The principles of atomic theory recognized today were conceived by
- Avogadro.
 - Bohr.
 - Dalton.
 - Rutherford.
- _____ 11. The total number of protons and neutrons in the nucleus of an atom is its
- atomic number.
 - Avogadro number.
 - mass number.
 - average atomic mass.
- _____ 12. All atoms of the same element have the same
- atomic mass.
 - number of neutrons.
 - mass number.
 - atomic number.
- _____ 13. Atoms of the same element can differ in
- chemical properties.
 - mass number.
 - atomic number.
 - number of protons and electrons.
- _____ 14. Chlorine has atomic number 17 and mass number 35. It has
- 17 protons, 17 electrons, and 18 neutrons.
 - 35 protons, 35 electrons, and 17 neutrons.
 - 17 protons, 17 electrons, and 52 neutrons.
 - 18 protons, 18 electrons, and 17 neutrons.
- _____ 15. Phosphorus-33 (atomic number 15) contains
- 33 protons.
 - 18 neutrons.
 - 33 neutrons.
 - 18 protons.
- _____ 16. Neon-22 contains 12 neutrons. It also contains
- 12 protons.
 - 22 protons.
 - 22 electrons.
 - 10 protons.
- _____ 17. Valence electrons determine an atom's

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Kingsley Augustine



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