

# Matter—Properties and Changes

## Section 3.1 Properties of Matter

In your textbook, read about physical properties and chemical properties of matter.

Use each of the terms below just once to complete the passage.

chemical  
density

mass  
properties

physical  
substance

Matter is anything with **(1)** \_\_\_\_\_ and volume. A **(2)** \_\_\_\_\_ is a form of matter with a uniform and unchanging composition. Substances have specific, unchanging **(3)** \_\_\_\_\_ that can be observed. Substances have both physical and chemical properties. **(4)** \_\_\_\_\_ properties can be observed without changing a substance's chemical composition. Color, hardness, and **(5)** \_\_\_\_\_ are examples. Other properties cannot be observed without changing the composition of a substance. These are called **(6)** \_\_\_\_\_ properties. An example is the tendency of iron to form rust when exposed to air.

Label each property as either *physical* or *chemical*.

- \_\_\_\_\_ 7. Chemical formula  $H_2O$
- \_\_\_\_\_ 8. Forms green carbonate when exposed to moist air
- \_\_\_\_\_ 9. Remains unchanged when in the presence of nitrogen
- \_\_\_\_\_ 10. Colorless
- \_\_\_\_\_ 11. Solid at normal temperatures and pressures
- \_\_\_\_\_ 12. Ability to combine with another substance
- \_\_\_\_\_ 13. Melting point
- \_\_\_\_\_ 14. Liquid at normal temperatures and pressures
- \_\_\_\_\_ 15. Boiling point is  $100^{\circ}C$
- \_\_\_\_\_ 16. Conducts electricity
- \_\_\_\_\_ 17. Density is  $\frac{1g}{cm^3}$

# Chemistry Study Guide Packet Answer

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Smith, Jack Price**



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## **Table of Contents Chemistry Study Guide Packet Answer**

1. Understanding the eBook Chemistry Study Guide Packet Answer
  - The Rise of Digital Reading Chemistry Study Guide Packet Answer
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Study Guide Packet Answer
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemistry Study Guide Packet Answer
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Study Guide Packet Answer
  - Personalized Recommendations
  - Chemistry Study Guide Packet Answer User Reviews and Ratings
  - Chemistry Study Guide Packet Answer and Bestseller Lists
5. Accessing Chemistry Study Guide Packet Answer Free and Paid eBooks
  - Chemistry Study Guide Packet Answer Public Domain eBooks
  - Chemistry Study Guide Packet Answer eBook Subscription Services
  - Chemistry Study Guide Packet Answer Budget-Friendly Options
6. Navigating Chemistry Study Guide Packet Answer eBook Formats
  - ePub, PDF, MOBI, and More
  - Chemistry Study Guide Packet Answer Compatibility with Devices
  - Chemistry Study Guide Packet Answer Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chemistry Study Guide Packet Answer
  - Highlighting and Note-Taking Chemistry Study Guide Packet Answer
  - Interactive Elements Chemistry Study Guide Packet Answer

8. Staying Engaged with Chemistry Study Guide Packet Answer
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chemistry Study Guide Packet Answer
9. Balancing eBooks and Physical Books Chemistry Study Guide Packet Answer
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemistry Study Guide Packet Answer
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Study Guide Packet Answer
  - Setting Reading Goals Chemistry Study Guide Packet Answer
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Study Guide Packet Answer
  - Fact-Checking eBook Content of Chemistry Study Guide Packet Answer
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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