

what elements of dirt is responsible for creating a healthy environment?

Introduction

The minerals in dirt are essential for creating a healthy environment for plants and animals. Minerals in dirt are like the food for our humans. The right dirt for example contains lots of calcium which creates a strong and healthy person, it's the same for plants. Plants needing sulfate as their humans need protein for building blocks for repairing and building. The sulfate supports creation of plant proteins. So if there is no sulfate the will be less plants, less plant means less pollutants. Sulfuric pollution removes globules becoming more water more water more drought. This may lead to a green house but it will affect people stopping the plant from growing. It can be said that more minerals in the dirt the better for all substances of earth. Ammonium sulfate is used to treat plants to build buildings, cars and farming and much more. $\text{Na}_2\text{SO}_4 + \text{Ca}(\text{OH})_2 \rightarrow \text{CaSO}_4 + 2\text{NaOH}$

Aim

To find the percentage of sulfate within the samples of dirt with the equation: $\text{Ba} + \text{SO}_4 \rightarrow \text{BaSO}_4$ to find the values of the sulfate with

Hypothesis

The dirt with the darker and the highest mineral is found in a higher biodiversity environment because the samples were collected will have the high sulfate levels. So sample 1 is the dirt with the highest mineral and the darkest color.

Sample 1 is from a moist forest like conditions in Diamond

plenty of plants around

Sample 2 is from a dry forest like conditions in known district.

tree mostly dead

Location and time

2021

10:45 am to 11:45 AM

- Materials and equipment
- Calculator
- 100ml beaker
- 1.0g
- 10ml of hydrochloric acid
- 25ml
- 25ml of 2.5% aqueous ammonia solution
- disinfectant water
- beakers 250mL, graduated
- burner
- range funnel
- 10ml measuring funnel
- burner and burner stand
- paper
- small paper
- paper funnel
- 100ml graduated flask
- Safety equipment (safety coat and safety glasses)
- Marker and pencil
- Measuring rod
- Large white tile
- Burner burner
- Heat probe mat
- glove mat
- tray stand
- pan
- weights

Risk assessment

when handling:

hydrochloric acid makes sure you pour slowly of hydrochloric acid into water combined with skin or clothing, wear a location not contact

burner chloride is harmful to health so wear a mask

After burner burner is hot don't touch with bare hands

the larger burner is hot after the heat don't use hands to move

when safety glasses and laboratory coat

variables

Independent variables

barium

platinum

amount of dirt sample

hydrochloric acid

Dependent variables

sulfate

disinfectant water

RESULTS

	amount of disinfectant water	amount of water recycled	amount of disinfectant	amount of sulfate	amount of dirt sample	percentage of sulfate in sulfate
1st	0.75g	0.75g	0.25g	0.0045g	0.0035g	1.00%
2nd	0.82g	0.82g	0.26g	0.0050g	0.0032g	1.56%

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