

General Chemistry II Chapter 15 Study Guide

- $S_{\text{gas}} > S_{\text{liquid}} > S_{\text{solid}}$
- For similar molecules, S° tends to increase with:
 - Increasing molar mass
 - Increased number of atoms in the formula

Standard entropy of a reaction ($\Delta S_{\text{rxn}}^\circ$):

- $\Delta S_{\text{rxn}}^\circ$: is the entropy change that occurs when all reactants and products are in their standard states:

$$\Delta S_{\text{rxn}}^\circ = \sum m S_{\text{products}}^\circ - \sum n S_{\text{reactants}}^\circ$$

where m and n are the mols of products and reactants, given by the coefficients in the balanced equation.

Suppose a reaction: $a \text{ A} + b \text{ B} \rightarrow c \text{ C} + d \text{ D}$

$$\sum m S_{\text{products}}^\circ = c \Delta S^\circ (\text{C}) + d \Delta S^\circ (\text{D})$$

$$\sum n S_{\text{reactants}}^\circ = a \Delta S^\circ (\text{A}) + b \Delta S^\circ (\text{B})$$

Remember to multiply ΔH° of each substance by its coefficient.

Predicting the sign of $\Delta S_{\text{rxn}}^\circ$:

- We can often predict the sign of $\Delta S_{\text{rxn}}^\circ$ for processes that involve a change in the number of moles of gas
 - $\Delta S_{\text{rxn}}^\circ$ is positive if the amount of gas increases
 - $\Delta S_{\text{rxn}}^\circ$ is negative if the amount of gas decreases
 - $\Delta S_{\text{rxn}}^\circ$ is likely to be positive if a new structure forms that has more freedom of motion

Gibbs free energy:

- Gibbs free energy (G): combines the enthalpy and entropy of a system:
 - $G = H - TS$
- Gibbs free energy change (ΔG): is a measure of the spontaneity of a process and of the useful energy available from it:
 - $\Delta G_{\text{rxn}} = \Delta H_{\text{rxn}} - T\Delta S_{\text{rxn}}$
- $\Delta G < 0$ for a spontaneous reaction ($-\Delta G$)
- $\Delta G > 0$ for a nonspontaneous reaction ($+\Delta G$)
- $\Delta G = 0$ for a process at equilibrium ($\Delta G = 0$)

Chemistry Chapter 15 Study Guide Answers

Martha Mackin



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Wastewater Operator Certification Study Guide John Giorgi, 2011-01-12 Student Study Guide and Solutions Manual to accompany General Organic and Biological Chemistry, 1e Kenneth W. Raymond, 2005-10-07 Finally readers have a shorter less intimidating introduction to general organic and biological chemistry Not only is Raymond's text concise it also takes an integrated approach to presenting important topics in a way that makes the material easier to understand In this approach similarities can be exploited and concepts reinforced The result is that readers see the strong connections that exist between these three branches of chemistry **Study Guide to Accompany Basics for Chemistry** Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students **Operator Certification Study Guide** John Giorgi, 2011-01-12 This book is a revision of the popular study guide for water system last published in 1993 This study resource is a practical tool for treatment plant operators and distribution system personnel as they prepare for the certification exam Actually formatting is used with the sample questions all of which have been reviewed by ABC Association of Board of Certification and are based on information contained in the WSO training series Water Treatment Textbook and the Water Distributor Operation Handbook Math formulas conversion factors and other resource references are also included Previous edition 0 89867 685 1 Study Guide with Answers to Selected Problems H. Stephen Stoker, Danny V. White, 2003-04 This supplement includes for each chapter a brief overview activities and practice problems to reinforce skills and a practice test The answers section includes answers for all odd numbered end of chapter exercises **Study Guide**

and Solutions Manual for McMurry's Organic Chemistry, Fifth Edition Susan McMurry, 2000 Provides answers and explanations to all in text and end of chapter exercises Also includes summaries of name reactions functional group synthesis and reactions lists of reagents and abbreviations and articles on topics ranging from infrared absorption frequencies to the Nobel Prize winners in Chemistry This edition now includes all new artwork expanded in text problems summary quizzes approximately every three chapters more detailed explanations in solutions and chapter outlines *Student Study Guide Fundamentals of Chemistry by David E. Goldberg* Margaret G. Kimble, 1994 **Student Study Guide to Accompany Petrucci's General Chemistry, 3rd. Ed** Robert K. Wismer, 1982 **Chemistry, Student Study Guide** John A. Olmsted, Gregory M. Williams, 2002-01-04 This third edition continues to innovate by providing students with an integrated and modern approach to the subject The text emphasizes the modern tools of chemistry while incorporating historical evidence and its unique molecular quantitative emphasis is further reinforced by an integrated media package developed by the authors Also of benefit is the just in time presentation of key content only providing details once they are needed While key topics and analytical techniques have been updated there is now an additional third chapter on chemical equilibrium The authors have also developed an expanded and more integrated problem solving emphasis that now incorporates a 4 step strategy throughout complete with text icons The whole is backed by a range of supplements including a new illustration program a tutorial CD interactive learningware an extensive Web CT component an instructor's resource CD and a solution CD **General Chemistry Study Guide Sixth Edition** Darrell D. Ebbing, 1999 *Study Guide for Chemistry by Steven S. Zumdahl* Martha B. Barrett, 1986 **Know Your 'O' Level Chemistry - A Study Guide , Study Guide and Solutions Book** Harold Hart, 1983 Study Guide and Solutions Manual, Fundamentals of General, Organic, and Biological Chemistry, Third Edition John McMurry, Susan McMurry, 1999 Provides worked out solutions to text problems along with chapter by chapter outlines and a variety of self tests at the end of each chapter **Study Guide with Solutions for Organic Chemistry, Third Edition** Ralph J. Fessenden, Joan S. Fessenden, 1986 Provides a set of additional drill problems chapter by chapter discussions and supplemental instructional material to help students master organic chemistry problem solving techniques *CliffsNotes AP Chemistry* Bobrow Test Preparation Services, 2009-02-09 The book itself contains chapter length subject reviews on every subject tested on the AP Chemistry exam as well as both sample multiple choice and free response questions at each chapter's end Two full length practice tests with detailed answer explanations are included in the book **Study Guide for Chemistry, Third Edition [by] Steven S. Zumdahl** Paul B. Kelter, 1993 Organic Chemistry, Study Guide T. W. Graham Solomons, 1996 This supplement accompanies an updated text which features more than 350 new problems Carbonyl chemistry is now covered in consecutive chapters The concept of ionic reactions is consolidated before students move on to study radical reactions Instructor's Manual and Study Guide Answers for the Human Body in Health and Disease Barbara Janson Cohen, Memmler, 1996 General Chemistry Darrell D. Ebbing, Larry K. Krannich, Joan I.

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