

General Chemistry II Chapter 15 Study Guide

- $S'_{\text{gas}} > S'_{\text{liquid}} > S'_{\text{solid}}$
- For similar molecules, S' tends to increase with:
 - Increasing molar mass
 - Increased number of atoms in the formula

Standard entropy of a reaction ($\Delta S'_{\text{rxn}}$):

- $\Delta S'_{\text{rxn}}$: is the entropy change that occurs when all reactants and products are in their standard states:

$$\Delta S'_{\text{rxn}} = \Sigma m S'_{\text{products}} - \Sigma n S'_{\text{reactants}}$$

where m and n are the mols of products and reactants, given by the coefficients in the balanced equation.

Suppose a reaction: $a A + b B \rightarrow c C + d D$

$$\Sigma m S'_{\text{products}} = c \Delta S'_{\text{rxn}} (C) + d \Delta S'_{\text{rxn}} (D)$$

$$\Sigma n S'_{\text{reactants}} = a \Delta S'_{\text{rxn}} (A) + b \Delta S'_{\text{rxn}} (B)$$

Remember to multiply ΔH° of each substance by its coefficient.

Predicting the sign of $\Delta S'_{\text{rxn}}$:

- We can often predict the sign of $\Delta S'_{\text{rxn}}$ for processes that involve a change in the number of moles of gas
 - $\Delta S'_{\text{rxn}}$ is positive if the amount of gas increases
 - $\Delta S'_{\text{rxn}}$ is negative if the amount of gas decreases
 - $\Delta S'_{\text{rxn}}$ is likely to be positive if a new structure forms that has more freedom of motion

Gibbs free energy:

- Gibbs free energy (G): combines the enthalpy and entropy of a system:
 - $G = H - TS$
- Gibbs free energy change (ΔG): is a measure of the spontaneity of a process and of the useful energy available from it:
 - $\Delta G_{\text{rxn}} = \Delta H_{\text{rxn}} - T\Delta S_{\text{rxn}}$
- $\Delta G < 0$ for a spontaneous reaction ($-\Delta G$)
- $\Delta G > 0$ for a nonspontaneous reaction ($+\Delta G$)
- $\Delta G = 0$ for a process at equilibrium ($\Delta G = 0$)

Chemistry Chapter 15 Study Guide Answers

Martha Mackin

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[Wastewater Operator Certification Study Guide](#) John Giorgi,2011-01-12 [Student Study Guide and Solutions Manual to accompany General Organic and Biological Chemistry, 1e](#) Kenneth W. Raymond,2005-10-07 Finally readers have a shorter less intimidating introduction to general organic and biological chemistry Not only is Raymond's text concise it also takes an integrated approach to presenting important topics in a way that makes the material easier to understand In this approach similarities can be exploited and concepts reinforced The result is that readers see the strong connections that exist between these three branches of chemistry

Study Guide to Accompany Basics for Chemistry Martha Mackin,2012-12-02

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Operator Certification Study Guide John

Giorgi,2011-01-12 This book is a revision of the popular study guide for water system last published in 1993 This study resource is a practical tool for treatment plant operators and distribution system personnel as they prepare for the certification exam Actually formatting is used with the sample questions all of which have been reviewed by ABC Association of Board of Certification and are based on information contained in the WSO training series Water Treatment Textbook and the Water Distributor Operation Handbook Math formulas conversion factors and other resource references are also included Previous edition 0 89867 685 1 [Study Guide with Answers to Selected Problems](#) H. Stephen Stoker,Danny V. White,2003-04 This supplement includes for each chapter a brief overview activities and practice problems to reinforce skills and a practice test The answers section includes answers for all odd numbered end of chapter exercises

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Study Guide with Solutions for Organic Chemistry, Third Edition Ralph J. Fessenden,Joan S. Fessenden,1986 Provides a set of additional drill problems chapter by chapter discussions and supplemental instructional material to help students master organic chemistry problem solving techniques *CliffsNotes AP Chemistry* Bobrow Test Preparation Services,2009-02-09 The book itself contains chapter length subject reviews on every subject tested on the AP Chemistry exam as well as both sample multiple choice and free response questions at each chapter s end Two full length practice tests with detailed answer explanations are included in the book

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