

Chemistry Applications

Study Guide

Define chemical reaction

A Chemical Reaction is a process where two or more substances interact with each other, causing their atoms to rearrange and form new substances with different properties

Define Physical Change:

Physical Change is when matter changes forms but not chemical identity

Give examples of physical changes

- Change of State (Solid, Liquid, Gas)
- Dissolving (Sugar in Water)
- Changing the Size or Shape of an Object (Crushing a Can)

Define Chemical Change:

During a Chemical Change a Chemical Reaction Occurs, New Substances Form, Chemical Reactions Produce new Substances by Changing the way Atoms are Arranged

Give examples of chemical changes:

- Rusting (The Statue of Liberty)
- Burning (A Campfire)

Define reactants and products:

Reactants:
What you start with, bonds are broken
Products:
What you end with, atoms are rearranged

List six types of evidence for Chemical Reactions

- Color Change
- Light Emission
- Formation of a Gas

Chemistry Concepts Applications Study Guide Answer

Sebastian Brünink



Chemistry Concepts Applications Study Guide Answer:

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