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Cbse Chemistry Lab Manual 2012 Class 1

Rufus Phillips Williams



Cbse Chemistry Lab Manual 2012 Class 1:

Practical/Laboratory Manual Chemistry Class - XI Er. Meera Goyal, 2021-05-29

1 Basic Laboratory Techniques

1 To cut a glass tube or glass rod
2 To bend the glass rod at an angle
3 To draw a glass jet from a glass tube
4 To bore a cork and fit a glass tube into it
Viva Voce

2 Characterisation and Purification of Chemical Substances

1 To determine the melting point of the given unknown organic compound and its identification simple laboratory technique
Viva Voce

2 To determine the boiling point of a given liquid when available in small quantity simple laboratory method
Viva Voce

3 To prepare crystals of pure potash alum $K_2SO_4 \cdot Al_2(SO_4)_3 \cdot 24H_2O$ from the given impure sample
4 To prepare the pure crystals of copper sulphate from the given crude sample
5 To prepare pure crystals of benzoic acid from a given impure sample
Viva Voce

3 Measurement of pH Values

1 To determine the pH value of vegetable juices fruit juices tap water and washing soda by using universal pH paper
2 To determine and compare the pH values of solutions of strong acid HCl and weak acid CH_3COOH of same concentration
3 To study the pH change in the titration of strong base Vs strong acid by using universal indicator paper
4 To study the pH change by common ion CH_3COO^- ion in case of weak acid CH_3COOH
5 To determine the change in pH value of weak base NH_4OH in presence of a common ion NH_4^+
Viva Voce

4 Chemical Equilibrium

1 To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions
2 To study the shift in equilibrium between $Co(H_2O)_6^{2+}$ and Cl^- ions by changing the concentrations of either of the ions
Viva Voce

5 Quantitative Analysis

1 To prepare M 10 oxalic acid solution by direct weighing method
2 To prepare M 10 solution of sodium carbonate by direct weighing method
3 To determine the strength of given solution of sodium hydroxide by titrating it against N 10 or M 20 solution of oxalic acid
4 To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N 10 or M 20 sodium carbonate solution
Viva Voce

6 Qualitative Analysis

Analysis of Anions
Analysis of Cations
Viva Voce

7 Detection of Elements in Organic Compounds

1 To detect the presence of nitrogen sulphur and halogens in a given organic compound by Lassaigne's test
2 To detect the presence of nitrogen sulphur and halogens in the given organic compound sample number by Lassaigne's test
Viva Voce

INVESTIGATORY PROJECTS

1 Checking of Bacterial Contamination in Water

1 To check the bacterial contamination in drinking water by testing sulphide ions
Viva Voce

2 Methods of Water Purification

1 To purify water from suspended impurities by using sedimentation
2 To purify water by boiling
3 To purify water by distillation method
4 To purify water by reverse osmosis technique
5 To purify water by GAC method
6 To purify water by bleach treatment
7 To purify water by oxidising agent
8 To purify water by ozone treatment method
Viva Voce

3 Water Analysis

1 To test the hardness of different water samples
Viva Voce

4 Foaming Capacity of Various Soaps

1 To compare the foaming capacity of different washing soaps
2 To study the effect of addition of sodium carbonate on foaming capacity of washing soap
Viva Voce

5 Tea Analysis

1 To study the acidity of different samples of tea leaves tea by using pH paper
Viva Voce

6 Analysis of Fruits and Vegetable Juices

1 To analysis the fruit and vegetable juices for the constituent present in them

Viva Voce 7 Rate of Evaporation 1 To study the rate of evaporation of different liquids IViva Voce 8 Effect of Acids and Bases on Tensile Strength of Fibres 1 To compare the tensile strength of natural fibres and synthetic fibres 2 To study the effect of acids and bases on tensile strength of different fibres Viva Voce Principles of Chemistry I Lab Manual - SCI 123, Fall 2012 ,2012 **General Chemistry I** Steven Rowley,2009-07-22 **Chemistry Lab Manual** Neena Sinha, R Rangarajan, R P Manchanda, R K Gupta, Rajesh Kumar, Lab Manual **Chemistry 1 Lab Manual C.O.S. ,2016** General Chemistry ... Laboratory Manual and Note Book American School (Lansing, Ill.),1906 **Elementary practical chemistry** George S. Newth,1921 General Chemistry I Steven Rowley,2005-08-23 **Elementary Practical Chemistry** George S Newth,2017-04-30 Elementary Practical Chemistry A laboratory manual for use in organized science schools is an unchanged high quality reprint of the original edition of 1896 Hansebooks is editor of the literature on different topic areas such as research and science travel and expeditions cooking and nutrition medicine and other genres As a publisher we focus on the preservation of historical literature Many works of historical writers and scientists are available today as antiques only Hansebooks newly publishes these books and contributes to the preservation of literature which has become rare and historical knowledge for the future *A Laboratory Manual of General Chemistry* Leon Burr Richardson,Andrew Jackson Scarlett,1929 **Practical Chemistry (For B.Sc. I, II and III Year Students)** Pandey O.P. & Bajpai D.N. & Giri S.,1972 FOR B Sc I II III YEAR STUDENTS **Chemistry 1. Laboratory Manual** ,1984 CHEM 111 Lab Manual General Chemistry Christofield,2022-06-06 **Chemistry Stage 1 Laboratory Manual** University of Auckland. Department of Chemistry,1967 *Laboratory Manual of General Chemistry* Harry Nicholls Holmes,1921 **Introduction to Chemistry 1 (Chem 2A Lab Manual)** ,2024-06-26 Chemistry Peter T. Wassell,2009-08-10 *Laboratory Manual of General Chemistry* Rufus Phillips Williams,1894 Elementary practical chemistry G. S. Newth,1913 **INTRODUCTION TO CHEMISTRY LAB MANUAL. DONALD. VITARELLI SIEGEL (MICHAEL J.),2023**

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