



GAS LAWS: Demo/ Lab

Lesson Plan: What have these simulations and the gas laws shown us?

Aim: How are pressure, volume, and temperature related?

Agenda:

- Do Now - 5 minutes
- Pressure: Internal vs. external - 10 minutes
- Gas Laws - Boyle's and Charles's Law - 15 minutes
- Demo: Balloon in hot water - 5 minutes
- Summary: Gases and how they behave - 15 minutes
- Homework: Question on gas laws

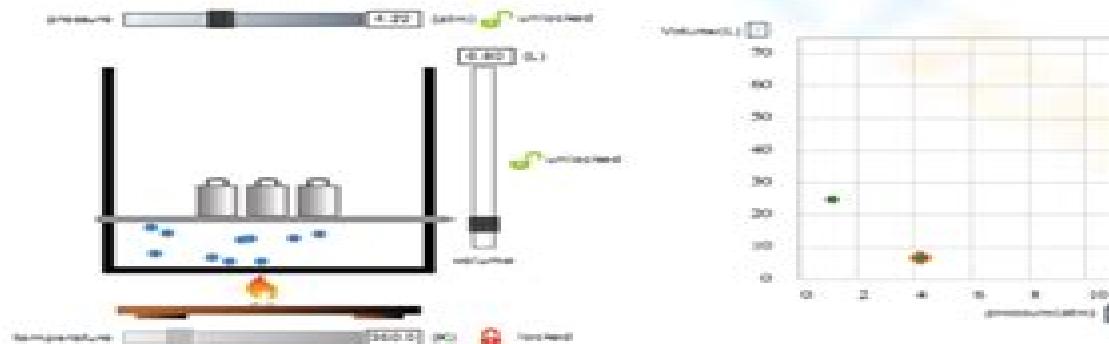
Pressure: Internal vs. external (start boiling water)

Draw a balloon... arrows going in and out. When something isn't moving, like a balloon, the pressure from the air all around us is equal to the amount the gas inside the balloon is pushing out.

So let's say I have a box that has a lid that slides up and down, if I apply pressure on the top and the lid stops sliding, that means that the gas inside the box is putting the same pressure on the inside to hold you there.

So if you apply pressure on the top of the box, do you see why pressure is going up inside the box as well?

If you apply external pressure, why doesn't the volume go to zero? What is holding it up? (no longer conforming to the Ideal Gas Law)



Demonstration Analogy

Use 8 volunteers... put them in a square. (4 inside [they are the gas molecules], 4 make the walls, I push them in) As the wall goes in what happens to the gas molecules? Do they like this?

Charles Law Computer Activity Answers

Scott C. Dulebohn

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Forensics in Chemistry Sara McCubbin, Angela Codron, 2012 Forensics seems to have the unique ability to maintain student interest and promote content learning. I still have students approach me from past years and ask about the forensics case and specific characters from the story I have never had a student come back to me and comment on that unit with the multiple choice test at the end from the Introduction to Forensics in Chemistry The Murder of Kirsten K. How did Kirsten K's body wind up at the bottom of a lake and what do wedding cake ingredients soil samples radioactive decay bone age blood stains bullet matching and drug lab evidence reveal about whodunit These mysteries are at the core of this teacher resource book which meets the unique needs of high school chemistry classes in a highly memorable way The book makes forensic evidence the foundation of a series of eight hands on week long labs As you weave the labs throughout the year and students solve the case the narrative provides vivid lessons in why chemistry concepts are relevant and how they connect All chapters include case information specific to each performance assessment and highlight the related national standards and chemistry content Chapters provide Teacher guides to help you set up Student performance assessments A suspect file to introduce the characters and new information about their relationships to the case Samples of student work that has been previously assessed and that serves as an answer key for you Grading rubrics Using Forensics in Chemistry as your guide you will gain the confidence to use inquiry based strategies and performance based assessments with a complex chemistry curriculum Your students may gain an interest in chemistry that rivals their fascination with Bones and CSI *ABA Journal*, 1966-08

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Computer Law Developments, 1984, 1985 **Environmental Law and Economics** Michael G. Faure, Roy A. Partain, 2019-10-10 A detailed overview of the law and economics methodology developed and employed by environmental lawyers and policymakers *Prentice Hall Chemistry*, 2000 *The Publishers Weekly*, 1972 **Current Law Index**, 2005

Illinois Chemistry Teacher, 1990-06 *The Software Encyclopedia*, 1988 **Current Index to Journals in Education**, 1999-10 *Books in Print Supplement*, 2002 *Foreign Companies in Hong Kong Yearbook*, 2007 **Datamation**, 1968

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