

**Chapter 10 Study Guide: The Mole****Matching**

Match each item with the correct statement below.

- a. molar volume
- b. molar mass
- c. atomic mass

- \_\_\_\_\_ 1. the number of grams of an element that is numerically equal to the atomic mass of the element in amu
- \_\_\_\_\_ 2. the mass of a mole of any element or compound
- \_\_\_\_\_ 3. the volume occupied by a mole of any gas at STP

Match each item with the correct statement below.

- |                            |                                      |
|----------------------------|--------------------------------------|
| a. representative particle | d. percent composition               |
| b. mole                    | e. standard temperature and pressure |
| c. Avogadro's number       | f. empirical formula                 |

- \_\_\_\_\_ 4. the number of representative particles of a substance present in 1 mole of that substance
- \_\_\_\_\_ 5. an atom, an ion, or a molecule, depending upon the way a substance commonly exists
- \_\_\_\_\_ 6. the SI unit used to measure amount of substance
- \_\_\_\_\_ 7. 0°C and 1 atm
- \_\_\_\_\_ 8. the percent by mass of each element in a compound
- \_\_\_\_\_ 9. the smallest whole number ratio of the atoms in a compound

**Multiple Choice**

Identify the choice that best completes the statement or answers the question.

- \_\_\_\_\_ 10. What SI unit is used to measure the number of representative particles in a substance?
  - a. kilogram
  - b. ampere
  - c. kelvin
  - d. mole
- \_\_\_\_\_ 11. How many hydrogen atoms are in 5 molecules of isopropyl alcohol,  $C_3H_7O$ ?
  - a.  $5 \times (6.02 \times 10^{23})$
  - b. 5
  - c. 15
  - d.  $2.5 \times (6.02 \times 10^{23})$
- \_\_\_\_\_ 12. Avogadro's number of representative particles is equal to one \_\_\_\_\_.
  - a. kilogram
  - b. gram
  - c. kelvin
  - d. mole

# Chapter 10 Study Guide The Mole Answer Key

**Alfred Marlyn Elliott, Robert Edward  
Hayes, John Ewald Lutz**



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