

Name _____ Date _____ Class _____

CHAPTER 15 STUDY GUIDE FOR CONTENT MASTERY

Solutions

Section 15.1 What are solutions?

In your textbook, read about the characteristics of solutions.

Use each of the terms below just once to complete the passage.

immiscible	liquid	solution	solute
insoluble	miscible	solute	solvent

Air is air! (1) solution of oxygen gas dissolved in nitrogen gas. The oxygen is air is the (2) solute and nitrogen is the (3) solvent. Because oxygen gas dissolves in a solvent, oxygen gas is air is (4) soluble substances. A substance that does not dissolve is (5) insoluble. (6) liquid solutions are the most common type of solutions. If one liquid is soluble in another liquid, such as acetic acid in water, the two liquids are (7) miscible. However, if one liquid is insoluble in another, the liquids are (8) immiscible.

Read about solvation in aqueous solutions in your textbook.

The diagram shows the hydration of solid sodium chloride to form an aqueous solution. Use the diagram to answer the following questions.



8. Hydration is solvation in which the solvent is water. What is solvation?

Solution is the process of surrounding solute particles with solvent particles to form a solution.

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CHAPTER 15 STUDY GUIDE FOR CONTENT MASTERY

Section 15.1 continued

10. As sodium chloride dissolves in water, what happens to the sodium and chloride ions?

The sodium and chloride ions are separated and surrounded by the water molecules.

11. Explain the orientation of the water molecules around the sodium ions and chloride ions.

Because the sodium ion is positively charged, it attracts the negatively charged portion of the water molecule (the oxygen atom) and repels the positively charged portion of the water molecule (the hydrogen atoms). Because the chloride ion is negatively charged, it attracts the positively charged portion of the water molecule and repels the negatively charged portion.

12. How does the strength of the attraction between water molecules and sodium and chloride ions compare with the strength of the attraction between the sodium ions and chloride ions? How do you know?

The attraction between the water molecules and the sodium and chloride ions is greater than the attraction between the sodium and chloride ions. The greater strength of attraction between the water molecules and the ions is what causes the solvation process to occur.

13. List three ways that the rate of solvation may be increased.

stirring or shaking the solution, breaking the solute into smaller pieces, and heating the solvent

In your textbook, read about heat of solution, solubility, and factors that affect solubility.

For each statement below, write true or false.

true 14. The overall energy change that occurs when a solution forms is called the heat of solution.

false 15. Solubility is a measure of the maximum amount of solute that dissolves in a given amount of solvent at a specified temperature and pressure.

false 16. Solvation continues as long as the solvation rate is less than the crystallization rate.

true 17. In a saturated solution, solvation and crystallization are in equilibrium.

true 18. Additional solute can be dissolved in an unsaturated solution.

true 19. The solubility of a gas dissolved in a liquid decreases as the temperature of the solution increases.

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Wayne S. Wooden, Randy Blazak



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Basic Pharmacology for Nurses16 Bruce D. Clayton,Michelle Willihnganz,2013-01-01 Basic Pharmacology for Nurses is a modern classic nursing pharmacology textbook known for its impeccably accurate drug content and its practical applications of the nursing process A consistent emphasis on health promotion through monitoring and patient education is a hallmark of the book Introductory units ground the reader in basic principles of pharmacology and medication administration Subsequent body system units apply the nursing process to every major disorder and appropriate nursing implications are discussed for every drug class to promote safe medication administration Provides an overview of relevant pathophysiology with an application of the nursing process in each drug chapter followed by nursing implications for each applicable drug class Covers the pharmacology of all major classes of drugs emphasizing side effects to expect side effects to report drug interactions and more Emphasizes patient education and health promotion with easy to recognize Patient Teaching boxes and reproducible Patient Self Assessment forms on the Evolve site Provides need to know coverage of herbal therapies and dietary supplements with an Herbal and Dietary Supplement Therapy chapter and Complementary and Alternative Therapies boxes throughout Increased emphasis on medication safety prepares nurses to practice more safely in a clinical environment that is both more pressured and more focused than ever on reducing drug errors The increased emphasis includes Chapter 7 Principles of Medication Administration do not confuse icons and an icon to signal ISMP s high alert drugs As the focus on pharmacology on the NCLEX continues to increase chapter ending Get Ready for the NCLEX R Examination sections include Key Points Additional Learning Resources and Review Questions for the NCLEX Examination Updated coverage of the most recent FDA approvals withdrawals and therapeutic uses provide impeccably accurate and current content to ensure safe drug therapy Updated and streamlined coverage of nursing care with a focus on the need to know material helps students focus on the most current need to know content *Questions and Answers to Help You Pass the Real Estate Appraisal Exam* Jeffrey D. Fisher,Dennis S. Tosh,2000 More than 200 000 appraisers are expected to take the real estate appraisal certification exam over the next few years Newly revised and updated this is the first authoritative exam guide to help appraisers prepare for the new state licensing requirements Basic Pharmacology for Nurses - E-Book Michelle Willihnganz,Bruce D. Clayton,2014-05-01 Basic Pharmacology for Nurses is a modern classic Known for its impeccably accurate drug content and practical applications of the nursing process this trusted text consistently emphasizes health promotion through monitoring and patient education Introductory units ground your knowledge in the basic principles of pharmacology and medication administration while subsequent body system units prepare you to apply the nursing process to every major disorder UNIQUE Two tiered application of the nursing process provides the general principles of care along with specific nursing implications for each drug class or individual drug Meticulously accurate and updated drug content keeps the book in step with the latest FDA approvals withdrawals and changes in therapeutic uses UNIQUE Introduction to

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Eureka Math Pre-K Study Guide Great Minds,2016-08-02 Eureka Math is a comprehensive content rich PreK 12 curriculum that follows the focus and coherence of the Common Core State Standards in Mathematics CCSSM and carefully sequences the mathematical progressions into expertly crafted instructional modules The companion Study Guides to Eureka Math gather the key components of the curriculum for each grade into a single location unpacking the standards in detail so that both users and non users of Eureka Math can benefit equally from the content presented Each of the Eureka Math Curriculum Study Guides includes narratives that provide educators with an overview of what students should be learning throughout the year information on alignment to the instructional shifts and the standards design of curricular components approaches to differentiated instruction and descriptions of mathematical models The Study Guides can serve as either a self study professional development resource or as the basis for a deep group study of the standards for a particular grade For teachers who are new to the classroom or the standards the Study Guides introduce them not only to Eureka Math but also to the content of the grade level in a way they will find manageable and useful Teachers familiar with the Eureka Math curriculum will also find this resource valuable as it allows for a meaningful study of the grade level content in a way that highlights the coherence between modules and topics The Study Guides allow teachers to obtain a firm grasp on what it is that students should master during the year The Eureka Math Curriculum Study Guide Grade PK provides an overview of all of the Pre Kindergarten modules including Counting to 5 Shapes Counting to 10 Comparison of Length Weight Capacity and Numbers to 5 and Addition and Subtraction Stories and Counting to 20 Introduction to Taxation Steven J. Rice,1994

Renegade Kids, Suburban Outlaws Wayne S. Wooden,Randy Blazak,2001 Topics covered include exploring boundary between deviance and criminality in the lives of young people who are deeply involved in the youth culture show how youth culture is not a set of categories so much as it is a dynamic and creative response to the confusions of growing up in modern society *Resources in Education* ,1991-04 *Business Communications* William C. Himstreet,Wayne Murlin Baty,Carol M. Lehman,1993 For an undergraduate or possibly graduate course in introductory business communication found in departments of business including marketing management and information systems business education English and communications *Psychology Study Guide* Richard B. Stalling,Wayne Weiten,Ronald E. Wasden,2006-02 By Richard Stalling and Ronald Wasden of Bradley University this study guide includes a review of key ideas people and terms complete

with fill in the blank short answer and matching questions This is followed by a self quiz made up of multiple choice questions
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