

Name _____ Date _____ Class _____

CHAPTER 15 STUDY GUIDE FOR CONTENT MASTERY

Solutions

Section 15.1 What are solutions?

In your textbook, read about the characteristics of solutions.

Use each of the terms below just once to complete the passage.

immiscible	liquid	soluble	solution
insoluble	miscible	solvent	solvent

Air is air! (1) solution of oxygen gas dissolved in nitrogen gas. The oxygen in air is the (2) solute and nitrogen is the (3) solvent. Because oxygen gas dissolves in a solvent, oxygen gas is air's (4) soluble substance. A substance that does not dissolve is (5) insoluble. (6) Liquid solutions are the most common type of solutions. If one liquid is soluble in another liquid, such as acetic acid in water, the two liquids are (7) miscible. However, if one liquid is insoluble in another, the liquids are (8) immiscible.

Read about solvation in aqueous solutions in your textbook.

The diagram shows the hydration of solid sodium chloride to form an aqueous solution. Use the diagram to answer the following questions.



8. Hydration is solvation in which the solvent is water. What is solvation?

Solution is the process of surrounding solute particles with solvent particles to form a solution.

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Chemistry: Matter and Change • Chapter 15

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Section 15.1 continued

10. As sodium chloride dissolves in water, what happens to the sodium and chloride ions?

The sodium and chloride ions are separated and surrounded by the water molecules.

11. Explain the attraction of the water molecules around the sodium ions and chloride ions.

Because the sodium ion is positively charged, it attracts the negatively charged portion of the water molecule (the oxygen atom) and repels the positively charged portion of the water molecule (the hydrogen atoms). Because the chloride ion is negatively charged, it attracts the positively charged portion of the water molecule and repels the negatively charged portion.

12. How does the strength of the attraction between water molecules and sodium and chloride ions compare with the strength of the attraction between the sodium ions and chloride ions? How do you know?

The attraction between the water molecules and the sodium and chloride ions is greater than the attraction between the sodium and chloride ions. The greater strength of attraction between the water molecules and the ions is what causes the solvation process to occur.

13. List three ways that the rate of solvation may be increased.

stirring or shaking the solution, breaking the solute into smaller pieces, and heating the solvent

In your textbook, read about heat of solution, solubility, and factors that affect solubility.

For each statement below, write true or false.

True _____ 14. The overall energy change that occurs when a solution forms is called the heat of solution.

False _____ 15. Solubility is a measure of the maximum amount of solute that dissolves in a given amount of solvent at a specified temperature and pressure.

False _____ 16. Solvation continues as long as the solvation rate is less than the crystallization rate.

True _____ 17. In a saturated solution, solvation and crystallization are in equilibrium.

True _____ 18. Additional solute can be dissolved in an unsaturated solution.

True _____ 19. The solubility of a gas dissolved in a liquid decreases as the temperature of the solution increases.

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Chapter 26 Study Guide For Content Mastery Answer Key

Carolyn J. Meyer



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Human Anatomy & Physiology Eldra Pearl Solomon,P. William Davis,1983

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Exploring Management John R. Schermerhorn, Jr.,Daniel G. Bachrach,2017-10-16 Exploring Management supports teaching and learning of core management concepts by presenting material in a straightforward conversational style with a strong emphasis on application With a focus on currency high interest examples and pedagogy that encourages critical thinking and personal reflection Exploring Management is the perfect balance between what students need and what instructors want

Introduction to Psychology Dennis Coon,1986

Study Guide George Karleskint,1991

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featuring specific clinical situations as well as critical thinking questions *Heart Sounds and Murmurs* Barbara Erickson,1991 Presents the information necessary to understand and recognize normal heart sounds and the most common abnormal sounds Tape was made with a heart sound simulator using studio recording equipment **Sociology** Joan Ferrante-Wallace,2006 **Im/Tb Cult Anthropology** Gary Ferraro,2005-04 **Renegade Kids, Suburban Outlaws** Wayne S. Wooden,Randy Blazak,2001 Topics covered include exploring boundary between deviance and criminality in the lives of young people who are deeply involved in the youth culture show how youth culture is not a set of categories so much as it is a dynamic and creative response to the confusions of growing up in modern society **Psychology Study Guide** Richard B. Stalling,Wayne Weiten,Ronald E. Wasden,2006-02 By Richard Stalling and Ronald Wasden of Bradley University this study guide includes a review of key ideas people and terms complete with fill in the blank short answer and matching questions This is followed by a self quiz made up of multiple choice questions The Study Guide is organized around learning objectives those same learning objectives that are included in the test bank **Study Guide, Development Across the Life Span, Second Edition, Robert S. Feldman** Carolyn J. Meyer,2000 **Christian Home Educators' Curriculum Manual** Cathy Duffy,2000 Reviews goal setting what to teach learning styles how to teach planning and record keeping resource addresses Cover *Essentials of Ecology* George Tyler Miller,2006-09 G Tyler Miller s worldwide bestsellers have evolved right along with the changing needs of your diverse student population Focused specifically on energizing and engaging all your students Miller and new contributor Scott Spoolman have been at work scrutinizing every line enhancing clarifying and streamlining to reduce word density as well as updating with the very latest environmental news and research The resulting texts are shorter clearer and so engaging that your students will actually want to read their assignments The ideal alternative to ecology texts that tend to be too difficult for non majors this succinct 13 chapter full color textbook covers scientific principles and concepts ecosystems evolution biodiversity population ecology and more New to this edition for instructors is PowerLecture a one stop shop for lecture prep that includes everything you need to create dynamic lectures all in one place **Mastering Psychology and Life** Karl Alfred Minke,1979

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