

Chapter 1: Introduction to Chemistry

What is chemistry? The study of matter and its properties.

What is a chemical reaction? A process in which one or more substances are converted into one or more new substances.

- | Properties | Characteristics |
|-------------------|------------------|
| 1. Mass | 1. Color |
| 2. Volume | 2. Shape |
| 3. Density | 3. State |
| 4. Boiling point | 4. Melting point |
| 5. Freezing point | 5. Solubility |
| 6. Conductivity | 6. Reactivity |
| 7. Flammability | 7. Toxicity |
| 8. Corrosiveness | 8. Radioactivity |
| 9. Hardness | 9. Brittleness |
| 10. Malleability | 10. Ductility |

What is the scientific method? A systematic approach to solving problems.

What is a hypothesis? A proposed explanation for a phenomenon.

What is an experiment? A procedure carried out to test a hypothesis.

What is a theory? A well-substantiated explanation of some aspect of the natural world.

What is a law? A statement of a principle or relationship that is based on repeated experimental observations.

What is a model? A representation of an object or system.

Property	Unit
Mass	kg
Volume	m ³
Density	kg/m ³
Boiling point	°C
Freezing point	°C
Solubility	g/100g
Conductivity	S/m
Flammability	Yes/No
Corrosiveness	Yes/No
Hardness	1-10
Malleability	Yes/No
Ductility	Yes/No
Radioactivity	Yes/No
Brittleness	Yes/No

Chapter 2: Matter and Its Properties

What is matter? Anything that has mass and takes up space.

1. Solid: Definite shape and volume.
2. Liquid: Indefinite shape, definite volume.
3. Gas: Indefinite shape and volume.
4. Plasma: Ionized gas.
5. Bose-Einstein condensate: A state of matter that occurs at very low temperatures.
6. Superfluid: A state of matter that flows without friction.
7. Supercritical fluid: A state of matter that is above its critical temperature and pressure.
8. Quark-gluon plasma: A state of matter that exists at extremely high temperatures.
9. Degenerate matter: A state of matter that exists at extremely high densities.
10. Dark matter: A hypothetical form of matter that is thought to make up most of the mass in the universe.

What is the difference between a pure substance and a mixture?

What is a chemical formula? A representation of a chemical compound.

What is a chemical equation? A representation of a chemical reaction.

What is a balanced chemical equation? A chemical equation in which the number of atoms of each element is the same on both sides.

What is a limiting reactant? A reactant that is completely consumed in a chemical reaction.

What is a theoretical yield? The maximum amount of product that can be formed from a given amount of reactants.

What is an actual yield? The amount of product that is actually obtained from a chemical reaction.

What is percent yield? The ratio of actual yield to theoretical yield, expressed as a percentage.

Chapter 3: Atomic Structure and Periodic Table

What is an atom? The smallest unit of matter that retains the chemical properties of an element.

1. Protons: Positively charged particles found in the nucleus.
2. Neutrons: Neutral particles found in the nucleus.
3. Electrons: Negatively charged particles found in the electron cloud.
4. Alpha particles: Helium nuclei consisting of two protons and two neutrons.
5. Beta particles: High-speed electrons or positrons.
6. Gamma rays: High-energy electromagnetic radiation.
7. X-rays: Electromagnetic radiation with shorter wavelengths than visible light.
8. Ultraviolet light: Electromagnetic radiation with shorter wavelengths than visible light.
9. Infrared light: Electromagnetic radiation with longer wavelengths than visible light.
10. Radio waves: Electromagnetic radiation with the longest wavelengths.

What is the periodic table? A table of elements organized by their chemical and physical properties.

What is a group? A vertical column of elements in the periodic table.

What is a period? A horizontal row of elements in the periodic table.

What is a metal? An element that is typically shiny, malleable, and a good conductor of electricity.

What is a nonmetal? An element that is typically dull, brittle, and a poor conductor of electricity.

What is a metalloid? An element that has properties of both metals and nonmetals.

What is a noble gas? A gas in Group 18 of the periodic table.

What is a halogen? A nonmetal in Group 17 of the periodic table.

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Ensheng Dong



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