

# General Chemistry Study Guide

## Chapters 1-3

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### Chapter 1

#### *Key Concepts:*

- 1. Chemistry is the study of matter and its properties.*
- 2. Matter is anything that has mass and occupies space.*
- 3. Mass is the amount of matter in an object.*
- 4. Volume is the amount of space an object occupies.*
- 5. Density is the mass of an object divided by its volume.*
- 6. Elements are the simplest form of matter and are made up of atoms.*
- 7. Atoms are the smallest particle of an element that has the chemical properties of that element.*
- 8. The periodic table is a list of the elements arranged in order of increasing atomic number.*
- 9. Compounds are molecules made up of two or more atoms of different elements.*
- 10. Mixtures are combinations of two or more substances that are not chemically bonded together.*

#### *Important Events:*

- 1. In 1687, Sir Isaac Newton published his law of gravity, which explained the force that attracts objects to one another.*
- 2. In 1766, Henry Cavendish discovered that water is made up of two elements, hydrogen and oxygen.*
- 3. In 1808, John Dalton proposed his atomic theory, which stated that all matter is made up of atoms.*

# College General Chemistry Study Guide

**Colorado Agricultural and Mechanical  
College. Department of  
Chemistry, Paul Reheard Frey**

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