

General Chemistry Study Guide

Chapters 1-3

Chapter 1

Key Concepts:

1. Chemistry is the study of matter and its properties.
2. Matter is anything that has mass and occupies space.
3. Mass is the amount of matter in an object.
4. Volume is the amount of space an object occupies.
5. Density is the mass of an object divided by its volume.
6. Elements are the simplest form of matter and are made up of atoms.
7. Atoms are the smallest particle of an element that has the chemical properties of that element.
8. The periodic table is a list of the elements arranged in order of increasing atomic number.
9. Compounds are molecules made up of two or more atoms of different elements.
10. Mixtures are combinations of two or more substances that are not chemically bonded together.

Important Events:

1. In 1687, Sir Isaac Newton published his law of gravity, which explained the force that attracts objects to one another.
2. In 1766, Henry Cavendish discovered that water is made up of two elements, hydrogen and oxygen.
3. In 1808, John Dalton proposed his atomic theory, which stated that all matter is made up of atoms.

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**Colorado Agricultural and Mechanical
College. Department of
Chemistry, Paul Reheard Frey**

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