



~~1000000000~~  $109 / 17.679 \times 100 = 56.6\%$



$$59 / 6.619 \times 100 = 75.6\%$$

5. a. TRUE  
b. FALSE  
c.  $\text{AgNO}_3 \rightarrow 169.91 \text{ g/mol}$   
d.  $6.02 \times 10^{23}$  particles = 1 mole of particles

# Ch 12 Stoichiometry Study Guide Answers

**James Joseph Gallagher**

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