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AP CHEMISTRY
Ch 14 – Acids and Bases

PRACTICE PROBLEMS

NATURE OF ACIDS AND BASES

Suggested Problems: 27, 29, 35, 37, 43, 47, 51, 55, 61, 69, 75, 79, 85, 95, 97, 103, 107, 111, 115, 121

28. Write the dissociation reaction and the corresponding K_a equilibrium expression for each of the following acids in water.

- a. HCN
- b. HOCH₃
- c. C₆H₅NH₃⁺

34. Use Table 14.2 to order the following from the strongest to the weakest base.



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Hobart Hurd Willard

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