

Concentration and Molarity III: Dilution and Evaporation PhET-Chemistry Labs

Introduction: In the first part of this lab, you learned about the actions of solid salts added to water to form solutions of various concentrations. In this exercise, you will use a powerful and simple formula to determine the concentration of a solution when it has been diluted with additional water or made more concentrated due to evaporation of water.



Concentration

Some handy vocabulary for you to define:

Molarity _____

Dilution _____

Evaporation _____

Procedure: PhET → Play with the Sims → Chemistry → Concentration Play Now!

$$M_1 V_1 = M_2 V_2$$

Part 1: Dilution

At this point you should be familiar with the Concentration simulation. Because some time may have passed, take a few minutes to re-learn the simulations. Pay particular attention to the effect of evaporation and addition of water to the solution's concentration.



How does the concentration change as additional water is added? _____

Why? _____



How does the concentration change as evaporation occurs? _____

Why is this? _____

Does evaporation change the concentration of a saturated solution? _____

Why is this? _____



Using the concentrated solution spigot, add a .25 Liter of Drink Mix to an empty beaker. What is the concentration? _____

Is this solution saturated? _____ How do you know? _____

Fill the beaker with another .25 Liter of water. What is the new concentration? _____

Complete the table below, using Cobalt chloride in an empty beaker, writing the concentration in the boxes provided.

Only .25L of spigot solution	.25L spigot + .25L water	.25L spigot + .50 L water	.25L spigot + .75 L water

Repeat the exercise, using Potassium dichromate in an empty beaker.

Only .25L of spigot solution	.25L spigot + .25L water	.25L spigot + .50 L water	.25L spigot + .75 L water

What do you notice about the concentration change as each addition of .25L of water is added to the concentrated spigot solution? _____

Concentration And Molarity Ii Dilution And Evaporation

**Robert Minard Garrels,Fred T.
Mackenzie,Cynthia Hunt**

Concentration And Molarity II Dilution And Evaporation:

Tietz Fundamentals of Clinical Chemistry and Molecular Diagnostics - E-Book Carl A. Burtis, David E. Bruns, 2014-02-26 A condensed easier to understand student version of the acclaimed Tietz Textbook of Clinical Chemistry and Molecular Diagnostics Tietz Fundamentals of Clinical Chemistry and Molecular Diagnostics 7th Edition uses a laboratory perspective in providing the clinical chemistry fundamentals you need to work in a real world clinical lab Coverage ranges from laboratory principles to analytical techniques and instrumentation analytes pathophysiology and more New content keeps you current with the latest developments in molecular diagnostics From highly respected clinical chemistry experts Carl Burtis and David Bruns this textbook shows how to select and perform diagnostic lab tests and accurately evaluate results Authoritative respected author team consists of two well known experts in the clinical chemistry world Coverage of analytical techniques and instrumentation includes optical techniques electrochemistry electrophoresis chromatography mass spectrometry enzymology immunochemical techniques microchips automation and point of care testing Learning objectives begin each chapter providing measurable outcomes to achieve after completing the material Key words are listed and defined at the beginning of each chapter and bolded in the text A glossary at the end of the book makes it quick and easy to look up definitions of key terms More than 500 illustrations plus easy to read tables help you understand and remember key concepts New chapters on molecular diagnostics include the principles of molecular biology nucleic acid techniques and applications and genomes and nucleic acid alterations reflecting the changes in this rapidly evolving field New content on clinical evaluation of methods kidney function tests and diabetes is added to this edition NEW multiple choice review questions at the end of each chapter allow you to measure your comprehension of the material NEW case studies on the Evolve companion website use real life scenarios to reinforce concepts

Chemistry Division Annual Progress Report for Period Ending

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ORNL , Journal of the Chemical Society ,1992 Radioactive Waste

Management U.S. Atomic Energy Commission, 1973 The Australian Journal of Experimental Biology and Medical Science ,1947 Henry's Clinical Diagnosis and Management by Laboratory Methods E-Book Richard A.

McPherson, Matthew R. Pincus, 2011-09-06 Recognized as the definitive book in laboratory medicine since 1908 Henry s Clinical Diagnosis and Management by Laboratory Methods edited by Richard A McPherson MD and Matthew R Pincus MD

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Chemical Biology, Selected Papers Of H G Khorana (With Introductions) H Gobind Khorana, 2000-05-24. The first two chapters of this invaluable book trace the developments of the chemistry and macromolecular structures respectively of proteins and nucleic acids. Similarly the introductions to the succeeding chapters review step by step the historical landmarks in the topics covered. These include discoveries of biological phosphate esters, nucleotides and nucleotide coenzymes important in intermediary metabolism, the nature of the genetic material and biological synthesis of proteins, formulation of the problem of the genetic code and perspectives on bioenergetics. The selected papers illustrate the developments of the chemical synthesis of nucleotides and nucleotide coenzymes of ribo and deoxy ribo polynucleotides, RNA, DNA, of the total synthesis of genes in the laboratory and principles for gene amplification, PCR. Another major section covers studies of enzymes that degrade nucleic acids, the structure of transfer RNA and its role in protein synthesis and the author's work on the elucidation of the genetic code. Finally there are descriptions of the studies on biological membranes and the membrane protein bacteriorhodopsin, a biological proton pump. These studies elucidated the mechanism of proton translocation which is central to bioenergetics.

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Chemical Cycles and the Global Environment Robert Minard Garrels, Fred T. Mackenzie, Cynthia Hunt, 1975

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Franklin Institute (Philadelphia, Pa.). Biochemical Research Foundation,1950 **Chemistry, a First Course** Jacqueline I. Kroschwitz,Melvin Winokur,1987 This proven introductory chemistry text has been thoroughly enhanced to prepare your students for a general chemistry or general organic and biological chemistry course With a logical organization and balanced treatment of concepts and practical applications Chemistry A First Course fosters a solid understanding of chemistry basics rather than just memorization of facts Throughout the text concepts are reinforced by referring to material previously discussed This respected author team s lively conversational and highly descriptive writing style will quickly engage your students and draw them into the world of chemistry **Reactor Chemistry Division Annual Progress Report for Period Ending** ... Oak Ridge National Laboratory. Reactor Chemistry Division,1960 Chemistry James C. Hill,2000 This text defines the concepts needed to learn or review cardiac auscultation The combination of audio and text explains how to identify and interpret normal and common abnormal heart sounds Some heart sounds are reproduced on a heart sound simulator allowing for a clear crisp grasp of specific individual sounds Others are recorded from real patients to distinguish between similar heart and lung sounds and to help the listener select the heart sounds from the auditory milieu *Canadian Journal of Chemistry* ,1976

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