

**Concentration and Molarity II: Dilution and Evaporation PhET-Chemistry Labs****Concentration**

**Introduction:** In the first part of this lab, you learned about the actions of solid salts added to water to form solutions of various concentrations. In this exercise, you will use a powerful and simple formula to determine the concentration of a solution when it has been diluted with additional water or made more concentrated due to evaporation of water.

**Some handy vocabulary for you to define:**

Molarity \_\_\_\_\_

Dilution \_\_\_\_\_

Evaporation \_\_\_\_\_

**Procedure:** PhET → Play with the Sims → Chemistry → Concentration Run Sim

$$M_1V_1 = M_2V_2$$

**Part I: Dilution**

At this point you should be familiar with the Concentration simulation. Because some time may have passed, take a few minutes to re-learn the simulations. Pay particular attention to the effect of evaporation and addition of water to the solution's concentration



How does the concentration change as additional water is added? \_\_\_\_\_

Why? \_\_\_\_\_



How does the concentration change as evaporation occurs? \_\_\_\_\_

Why is this? \_\_\_\_\_

Does evaporation change the concentration of a saturated solution? \_\_\_\_\_

Why is this? \_\_\_\_\_



Using the concentrated solution spigot, add a ½ Liter of Drink Mix to an empty beaker. What is the concentration?

\_\_\_\_\_ Is this solution saturated? \_\_\_\_\_ How do you know? \_\_\_\_\_

Fill the beaker with another ½ Liter of water. What is the new concentration? \_\_\_\_\_

Complete the table below, using Cobalt chloride in an empty beaker, writing the concentration in the boxes provided.

Only .25L of spigot solution	.25L spigot+.25L water	.25L spigot + .50 L water	.25L spigot + .75 L water

Repeat the exercise, using Potassium dichromate in an empty beaker.

Only .25L of spigot solution	.25L spigot+.25L water	.25L spigot + .50 L water	.25L spigot + .75 L water

What do you notice about the concentration change as each addition of .25L of water is added to the concentrated spigot solution? \_\_\_\_\_

\_\_\_\_\_

# Concentration And Molarity Ii Dilution And Evaporation

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## **Concentration And Molarity Ii Dilution And Evaporation:**

Tietz Fundamentals of Clinical Chemistry and Molecular Diagnostics - E-Book Carl A. Burtis, David E. Bruns, 2014-02-26 A condensed easier to understand student version of the acclaimed Tietz Textbook of Clinical Chemistry and Molecular Diagnostics Tietz Fundamentals of Clinical Chemistry and Molecular Diagnostics 7th Edition uses a laboratory perspective in providing the clinical chemistry fundamentals you need to work in a real world clinical lab Coverage ranges from laboratory principles to analytical techniques and instrumentation analytes pathophysiology and more New content keeps you current with the latest developments in molecular diagnostics From highly respected clinical chemistry experts Carl Burtis and David Bruns this textbook shows how to select and perform diagnostic lab tests and accurately evaluate results Authoritative respected author team consists of two well known experts in the clinical chemistry world Coverage of analytical techniques and instrumentation includes optical techniques electrochemistry electrophoresis chromatography mass spectrometry enzymology immunochemical techniques microchips automation and point of care testing Learning objectives begin each chapter providing measurable outcomes to achieve after completing the material Key words are listed and defined at the beginning of each chapter and bolded in the text A glossary at the end of the book makes it quick and easy to look up definitions of key terms More than 500 illustrations plus easy to read tables help you understand and remember key concepts New chapters on molecular diagnostics include the principles of molecular biology nucleic acid techniques and applications and genomes and nucleic acid alterations reflecting the changes in this rapidly evolving field New content on clinical evaluation of methods kidney function tests and diabetes is added to this edition NEW multiple choice review questions at the end of each chapter allow you to measure your comprehension of the material NEW case studies on the Evolve companion website use real life scenarios to reinforce concepts

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*Reports of the Biochemical Research Foundation of the Franklin Institute* Franklin Institute (Philadelphia, Pa.). Biochemical Research Foundation, 1951 Consists chiefly of reprints from various medical journals

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**Ending ...** Oak Ridge National Laboratory. Reactor Chemistry Division,1960      Chemistry James C. Hill,2000 This text  
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