

CHAPTER 1 THE LANGUAGE OF CHEMISTRY

Exercise 1(A)

Q.1. What is a symbol ? What is information does it convey?

Ans. Symbol :A symbol represent short form of an element.

It is convey following element.

- i) It represents a specific element.
- ii) It represents one atom of an element.
- iii) A symbol represents one atom of an element.
- iv) It represents no. of times an atom is heavier than one atomic mass unit (a.m.u) taken as a standard.

Q.2. Why is the symbol S for Sulphur, but Na for element represented by the Au,pb,Sn,Hg.

Ans. IUPAC means International Union of pure of and Applied Chemistry.

Au-Gold ,Pb-Lead, Sn-Tin, Hg – Mercury.

Q.4. IF the symbol for Cobalt, Co, were written a Co, what would be wrong with it ?

Ans. Co Stands for Cobalt, if we write it Co then it shows itself as a compound containing two non-metal i.e. Carbon and oxygen which form a gas carbon monoxide which is named by Greek Prefix.

Q.5. What do the following symbol for (a) H (b) H_2 (c) 2H (d) $2H_2$

Ans. (a)H stands for one atom of hydrogen.

(b) H_2 stands for one molecule of hydrogen.

(c) 2H Stands for one molecule of hydrogen.

(d) $2H_2$ Stands for two atoms and one molecule of hydrogen.

Q.6. What is meant by atomicity ? Name a diatomic element.

Ans. Atomicity : The number of atoms in a molecule of an element is called its atomicity.

Diatomic molecules – Element having two atoms in their molecules, e.g., hydrogen

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Table of Contents Concise Chemistry Icse Class 9

1. Understanding the eBook Concise Chemistry Icse Class 9
 - The Rise of Digital Reading Concise Chemistry Icse Class 9
 - Advantages of eBooks Over Traditional Books
2. Identifying Concise Chemistry Icse Class 9
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Concise Chemistry Icse Class 9
 - User-Friendly Interface
4. Exploring eBook Recommendations from Concise Chemistry Icse Class 9
 - Personalized Recommendations

- Concise Chemistry Icse Class 9 User Reviews and Ratings
- Concise Chemistry Icse Class 9 and Bestseller Lists
- 5. Accessing Concise Chemistry Icse Class 9 Free and Paid eBooks
 - Concise Chemistry Icse Class 9 Public Domain eBooks
 - Concise Chemistry Icse Class 9 eBook Subscription Services
 - Concise Chemistry Icse Class 9 Budget-Friendly Options
- 6. Navigating Concise Chemistry Icse Class 9 eBook Formats
 - ePub, PDF, MOBI, and More
 - Concise Chemistry Icse Class 9 Compatibility with Devices
 - Concise Chemistry Icse Class 9 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Concise Chemistry Icse Class 9
 - Highlighting and Note-Taking Concise Chemistry Icse Class 9
 - Interactive Elements Concise Chemistry Icse Class 9
- 8. Staying Engaged with Concise Chemistry Icse Class 9
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Concise Chemistry Icse Class 9
- 9. Balancing eBooks and Physical Books Concise Chemistry Icse Class 9
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Concise Chemistry Icse Class 9
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Concise Chemistry Icse Class 9
 - Setting Reading Goals Concise Chemistry Icse Class 9
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Concise Chemistry Icse Class 9
 - Fact-Checking eBook Content of Concise Chemistry Icse Class 9

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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