

Quiz & Worksheet - Coordination Compounds

1. What is the name of the compound $[\text{Fe}(\text{H}_2\text{O})_6]^+$?

- ☐ Hexaaquairon (III)
- ☐ Hexaaquairon
- ☐ Hexaaquairon (II)
- ☐ Hexaaquairon (0)

2. What is the name of the compound $[\text{CrCl}_4(\text{OH}_2)_2]^-$?

- ☐ diaquatetrachlorochromate (II) ion
- ☐ diaquatetrachlorochromate (III) ion
- ☐ diaquatetrachlorochromium (III) ion
- ☐ diaquachlorochromate (III) ion

3. Predict the number of unpaired electrons in $[\text{Ru}(\text{NH}_3)_6]^{3+}$

- ☐ 1
- ☐ 2
- ☐ 3
- ☐ 0

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Coordination Compounds Study Guide

Thomas Elliott Taylor



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buffers typically consist of an acid base pair with the acid and base differing by the presence or absence of a proton a conjugate acid base pair for instance one of the buffers that maintain the ph of human blood involves carbonic acid H_2CO_3 and its conjugate base the bicarbonate ion HCO_3^-

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