

ATOMIC STRUCTURE

RUTHERFORD'S MODEL

- Discovered that an atom is mostly made up of empty space
- Center of the atom is a small dense positively charged nucleus
- This nucleus holds the negatively charged particles in the atom

BOHR'S MODEL

- There are 2 main pieces of an atom:
- Nucleus
- Electrons

THE OLD MODEL OF THE ATOM



WAVE-MECHANICAL MODEL

- Electrons are moving so fast that they look like a cloud. Orbitals are regions of most likely electron location



WHAT IS TRUE TODAY...

- The nucleus is composed of 2 things:
 - PROTONS - positively charged particles
 - NEUTRONS - neutral charged particles
- Electrons are in a cloud - they are in orbitals
 - Electrons DO move around the nucleus in different patterns



Subatomic Particle	Location in the Atom	Charge	Weight
proton	center (in the nucleus)	positive	1 amu
neutron	center (in the nucleus)	neutral (no charge)	1 amu
electron	around the nucleus in clouds or orbitals	negative	Almost equal to 1/1836

IONS

- If an atom gains or loses an electron, it is called an ion
- This gives the atom a charge:
 - Lose an electron, the atom becomes positive
 - Gain an electron, the atom becomes negative

ISOTOPES

- An isotope is when an atom is physically the same
- An isotope is still the same element, it is just a little different from every other atom of the same element
 - slightly different mass

Chemistry Atomic Structure Study Guide

Nancy A. Sickles Corbett



Chemistry Atomic Structure Study Guide:

Hazmat Chemistry Study Guide (Second Edition) Jill Meryl Levy, 2011 Organic Chemistry, 12e Binder Ready Version Study Guide & Student Solutions Manual T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2016-04-11 This is the Student Study Guide Solutions Manual to accompany Organic Chemistry 12th Edition The 12th edition of Organic Chemistry continues Solomons Fryhle Snyder's tradition of excellence in teaching and preparing students for success in the organic classroom and beyond A central theme of the authors approach to organic chemistry is to emphasize the relationship between structure and reactivity To accomplish this the content is organized in a way that combines the most useful features of a functional group approach with one largely based on reaction mechanisms The authors philosophy is to emphasize mechanisms and their common aspects as often as possible and at the same time use the unifying features of functional groups as the basis for most chapters The structural aspects of the authors approach show students what organic chemistry is Mechanistic aspects of their approach show students how it works And wherever an opportunity arises the authors show students what it does in living systems and the physical world around us Study Guide for Chemistry, Third Edition [by] Steven S. Zumdahl Paul B. Kelter, 1993 **Resources in education** , 1982-02 **Chemistry Basics (Speedy Study Guide)** Speedy Publishing, 2014-12-28 A study guide of chemistry basics would give students bite sized chunks of information on a complex subject making the task of learning it easier and more enjoyable Study guides present detailed information on a topic in a compact and simplistic way which helps the student grasp more of what they are being taught Also the visual aids that usually are found in study guides can be very effective in keeping the attention of the child as well as embedding the material in their memory Study Guide/Selected Solutions Manual Julie R. Frentrop, 2002-06 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all odd numbered end of chapter problems in the text book **Barron's how to Prepare for College Entrance Examinations** Samuel C. Brownstein, Mitchel Weiner, 1974 A guide to preparing for college entrance examinations with emphasis on study programs for the verbal mathematics and standard written English parts of the SAT Includes practice tests **Student Study Guide to Accompany Essentials Anatomy and Physiology** Jackie L Butler, Shier, Ricki Lewis, Dr, Nancy Ann Sickles Corbett, 1999-06 **Student Study Guide, Human Anatomy and Physiology, Fifth Edition** Nancy A. Sickles Corbett, 1990 **A Guide to the Literature of Chemistry** Evan Jay Crane, Austin McDowell Patterson, 1927 **Chemical Abstracts** , 1926 McGraw-Hill Encyclopedia of Science and Technology Daniel N. Lapedes, 1977 Accompanied by Readers guide 33 p 28 cm New York McGraw Hill c1977 Call number Q121 M3 1977 Suppl 1 Accompanied by Study guide 50 p 28 cm New York McGraw Hill c1977 Call number Q121 M3 1977 Suppl 2 **Ohio Schools** , 1964 **The Guide to Nature** , 1922 University High School Journal , 1927 **Study Guide [for]** Charles Seiger, 2000 Pennsylvania School Journal , 1965 Includes Official program of the meeting of the Pennsylvania State Educational Association some times separately paged

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Table of Contents Chemistry Atomic Structure Study Guide

1. Understanding the eBook Chemistry Atomic Structure Study Guide
 - The Rise of Digital Reading Chemistry Atomic Structure Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Atomic Structure Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Atomic Structure Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Atomic Structure Study Guide

- Personalized Recommendations
- Chemistry Atomic Structure Study Guide User Reviews and Ratings
- Chemistry Atomic Structure Study Guide and Bestseller Lists
- 5. Accessing Chemistry Atomic Structure Study Guide Free and Paid eBooks
 - Chemistry Atomic Structure Study Guide Public Domain eBooks
 - Chemistry Atomic Structure Study Guide eBook Subscription Services
 - Chemistry Atomic Structure Study Guide Budget-Friendly Options
- 6. Navigating Chemistry Atomic Structure Study Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Atomic Structure Study Guide Compatibility with Devices
 - Chemistry Atomic Structure Study Guide Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Atomic Structure Study Guide
 - Highlighting and Note-Taking Chemistry Atomic Structure Study Guide
 - Interactive Elements Chemistry Atomic Structure Study Guide
- 8. Staying Engaged with Chemistry Atomic Structure Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Atomic Structure Study Guide
- 9. Balancing eBooks and Physical Books Chemistry Atomic Structure Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Atomic Structure Study Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Atomic Structure Study Guide
 - Setting Reading Goals Chemistry Atomic Structure Study Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Atomic Structure Study Guide

- Fact-Checking eBook Content of Chemistry Atomic Structure Study Guide
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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