

Study Guide for Chemistry EOC Assessment

Your test will consist of multiple-choice questions over the following topics.

1. Metric units for mass, length, volume
2. Metric to metric conversions (K H Da m/l/g D C M)
3. Know how to put numbers in scientific notation and take them out.
4. Significant Digit rules
5. Difference between chemical and physical properties and changes. There will be multiple questions on this.
6. Ways to tell if a chemical change (chemical reaction) is taking place.
7. Pure substances and mixtures
 - a. heterogeneous mixture
 - b. homogeneous mixture
8. Density & how it relates to floating and sinking
9. Use the equation for density
$$\text{Density} = \frac{\text{Mass}}{\text{volume}}$$

(g/ml) (g) (ml)
10. Know the phases of matter and the characteristics of each phase.
11. Law of Conservation of Matter
12. What is a mole in chemistry?
13. Convert grams to moles, and moles to grams
14. Define an atom
15. Structure of an atom and where each subatomic particle is located.
 - a. Nucleus (contains protons and neutrons)
 - b. Protons (positive)
 - c. Neutrons (neutral)
 - d. Electrons (negative)
16. Isotopes
17. Elements
18. If given atomic # and mass, find the number of protons and neutrons in an atom.
19. Find the number of valence electrons, and tell what it says about reactivity.
20. Periodic Table
 - a. Know about the families
 - b. Group vs. period
21. Know about the properties of metals and nonmetals.
 - a. Ductile, malleable, conductors
22. Ionic & Covalent Compounds

How do they form? (Transfer or gain electrons)

Characteristics of Each
23. Be able to name ionic compounds
24. Predict oxidation numbers
25. Chemical Equations:

Know the parts of an equation

5 main types of reactions-know about them and be able to identify them (Synthesis, Decomposition, Single Displacement, Double Displacement, Combustion)
26. Acids & Bases-

Macro and Microscopic properties of each

What ions do they form, how do they affect litmus, what is their pH range?

Identify examples of each- Mono, di, and polyprotic acids

Tell if they are strong or weak

Acid-base indicator

Chemistry Eoc Study Guide Answers

**Denise Guinn, Rebecca Brewer, Rachel
C. Lum**



Chemistry Eoc Study Guide Answers:

Cracking the Virginia SOL Paul Foglino, Princeton Review (Firm), 2000 The Princeton Review realizes that acing the Chemistry exam is very different from getting straight As in school The Princeton Review doesn't try to teach students everything there is to know about chemistry only the techniques they'll need to score higher on the exam There's a big difference In **Cracking the Virginia SOL EOC Chemistry** TPR will teach test takers how to think like the test makers and Score higher by mastering what test takers need to know with proven techniques Use word association to solve problems when unsure of the correct answers Ace the exam by understanding the logic of the test makers and applying it to solutions This book includes 2 full length simulated exams All of TPR's sample test questions are just like the ones test takers will see on the actual End of Course Chemistry exam and TPR fully explains every solution Contents Include The Mystery Exams About the SOL Exams How to Crack the Chemistry EOC II Chemistry Review Atomic Structure Anatomy of the Periodic Table Chemical Compounds Chemical Reactions Chemistry Math Gases Phase Changes Solutions Equilibrium Acids and Bases How Chemical Reactions Happen Laboratory III The Princeton Review Practice Tests **Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today** Seager/Slabaugh, 2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application **South Dakota Chemistry Eoc Success Strategies Study Guide** South Dakota Eoc Exam Secrets Test Prep, 2014-08-22 South Dakota Chemistry EOC Success Strategies helps you ace the South Dakota End of Course Exams without weeks and months of endless studying Our comprehensive South Dakota Chemistry EOC Success Strategies study guide is written by our exam experts who painstakingly researched every topic and concept that you need to know to ace your test Our original research reveals specific weaknesses that you can exploit to increase your exam score more than you've ever imagined South Dakota Chemistry EOC Success Strategies includes The 5 Secret Keys to South Dakota EOC Success Time is Your Greatest Enemy Guessing is Not Guesswork Practice Smarter Not Harder Prepare Don't Procrastinate Test Yourself A comprehensive General Strategy review including Make Predictions Answer the Question Benchmark Valid Information Avoid Fact Traps Milk the Question The Trap of Familiarity Eliminate Answers Tough Questions Brainstorm Read Carefully Face Value Prefixes Hedge Phrases Switchback Words New Information Time Management Contextual Clues Don't Panic Pace Yourself Answer Selection Check Your Work Beware of Directly Quoted Answers Slang Extreme Statements Answer Choice Families Along with a complete in depth study guide for your specific South Dakota EOC exam and much more **Roadmap to the Virginia SOL** Princeton Review, 2005 Roadmap to the Virginia SOL EOC Chemistry includes strategies that are proven to enhance student performance The experts at The

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Exemplary Science in Grades 9-12 Robert Eugene Yager,2005 Essay titles reveal the range of programs and creativity this book encompasses The book ends with a summary chapter on successes and continuing challenges in meeting the Standards visions for improving high school science **Books In Print 2004-2005** Ed Bowker Staff,Staff Bowker, Ed,2004

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