

ChemCom 1st Semester Study Guide

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- _____ 1. Which of the following observations would predict a physical change?
a. rusting of a nail
b. ignition of paper
c. emission of a gas
d. evaporation of water
- _____ 2. Which of these is a chemical property?
a. resistance to corrosion
b. malleability
c. boiling point
d. solubility in water
- _____ 3. Properties such as melting point, density, and electrical conductivity are examples of
a. chemical properties
b. nuclear properties
c. physical properties
d. electrical properties
- _____ 4. All of the following properties are physical properties of a substance except
a. boiling point
b. hardness
c. malleability
d. combustibility
- _____ 5. What two measurements of a substance make up the substance's density value?
a. mass and weight
b. volume and pressure
c. mass and volume
d. temperature and mass
- _____ 6. A physical property of a substance
a. can be determined without changing the identity of the substance
b. depends on the way a substance reacts
c. is a property that helps support life
d. involves breaking the chemical bonds that hold compounds together
- _____ 7. A change in which bonds are broken and new bonds are formed, resulting in new substances being produced is called
a. a phase change
b. a nuclear change
c. a chemical change
d. a physical change
- _____ 8. Which of these observations is evidence of a chemical change?
a. A drink mix dissolves in water.
b. Bubbles of a gas explode when ignited.
c. Rubbing alcohol evaporates from a surface.
d. A pop can is crushed by a hammer.
- _____ 9. All of the following are chemical changes except
a. Wood burns in a fireplace
b. A glass of milk becomes sour.
c. Mold "grows" on cheese.
d. Sugar dissolves in water.
- _____ 10. Which of these is a chemical change?
a. Tomato juice is made from ripe tomatoes.
b. A pencil is sharpened.
c. A brass doorknob tarnishes.
d. Friction wears away the tread on a tire after driving it many miles.
- _____ 11. The following statement contains five numbered properties of chlorine:
Chlorine is a yellow gas (1) that is more dense than air (2). It forms hydrochloric acid when breathed into the lungs (3) and kills bacteria (4) when dissolved in water (5).
How many of the five properties in the statement are physical properties?
a. 1
b. 2

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