

Anions		chloride	oxide	iodide	hydride
Cations	Anions	$\text{Cl}^-$	$\text{O}^{2-}$	$\text{I}^-$	$\text{H}^-$
Sodium	$\text{Na}^+$	$\text{NaCl}$	$\text{Na}_2\text{O}$	$\text{NaI}$	$\text{NaH}$
Potassium	$\text{K}^+$	$\text{KCl}$	$\text{K}_2\text{O}$	$\text{KI}$	$\text{KH}$
Magnesium	$\text{Mg}^{2+}$	$\text{MgCl}_2$	$\text{MgO}$	$\text{MgI}_2$	$\text{MgH}_2$
Calcium	$\text{Ca}^{2+}$	$\text{CaCl}_2$	$\text{CaO}$	$\text{CaI}_2$	$\text{CaH}_2$
Copper(II)	$\text{Cu}^{2+}$	$\text{CuCl}_2$	$\text{CuO}$	$\text{CuI}_2$	$\text{CuH}_2$
Iron(II)	$\text{Fe}^{2+}$	$\text{FeCl}_2$	$\text{FeO}$	$\text{FeI}_2$	$\text{FeH}_2$
Iron(III)	$\text{Fe}^{3+}$	$\text{FeCl}_3$	$\text{Fe}_2\text{O}_3$	$\text{FeI}_3$	$\text{FeH}_3$
Silver	$\text{Ag}^+$	$\text{AgCl}$	$\text{Ag}_2\text{O}$	$\text{AgI}$	$\text{AgH}$
Zinc	$\text{Zn}^{2+}$	$\text{ZnCl}_2$	$\text{ZnO}$	$\text{ZnI}_2$	$\text{ZnH}_2$
Aluminum	$\text{Al}^{3+}$	$\text{AlCl}_3$	$\text{Al}_2\text{O}_3$	$\text{AlI}_3$	$\text{AlH}_3$

### Set 2

Anions		bromide	oxide	fluoride	astatide
Cations	Anions	$\text{Br}^-$	$\text{O}^{2-}$	$\text{F}^-$	$\text{At}^-$
Lithium	$\text{Li}^+$	$\text{LiBr}$	$\text{Li}_2\text{O}$	$\text{LiF}$	$\text{LiAt}$
Barium	$\text{Ba}^{2+}$	$\text{BaBr}_2$	$\text{BaO}$	$\text{BaF}_2$	$\text{BaAt}_2$
Cesium	$\text{Cs}^+$	$\text{CsBr}$	$\text{Cs}_2\text{O}$	$\text{CsF}$	$\text{CsAt}$
Strontium	$\text{Sr}^{2+}$	$\text{SrBr}_2$	$\text{SrO}$	$\text{SrF}_2$	$\text{SrAt}_2$
Copper(I)	$\text{Cu}^+$	$\text{CuBr}$	$\text{Cu}_2\text{O}$	$\text{CuF}$	$\text{CuAt}$
Copper(II)	$\text{Cu}^{2+}$	$\text{CuBr}_2$	$\text{CuO}$	$\text{CuF}_2$	$\text{CuAt}_2$
Lead(II)	$\text{Pb}^{2+}$	$\text{PbBr}_2$	$\text{PbO}$	$\text{PbF}_2$	$\text{PbAt}_2$
Lead(IV)	$\text{Pb}^{4+}$	$\text{PbBr}_4$	$\text{PbO}_2$	$\text{PbF}_4$	$\text{PbAt}_4$
Gallium	$\text{Ga}^{3+}$	$\text{GaBr}_3$	$\text{Ga}_2\text{O}_3$	$\text{GaF}_3$	$\text{GaAt}_3$
Nickel(II)	$\text{Ni}^{2+}$	$\text{NiBr}_2$	$\text{NiO}$	$\text{NiF}_2$	$\text{NiAt}_2$

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**Martin Van Buren McGill**

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