

Chemical Reactions

Section 9.1 Reactions and Equations

In your textbook, read about evidence of chemical reactions.

For each statement, write *yes* if evidence of a chemical reaction is present. Write *no* if there is no evidence of a chemical reaction.

- | | |
|---------------|---|
| <u>yes</u> | 1. A tomato smells rotten. |
| <u>no</u> | 2. A drinking glass breaks into smaller pieces. |
| <u>no</u> | 3. A piece of ice melts. |
| <u>yes</u> | 4. Drain cleaner is mixed with water and the solution becomes warm. |
| <u>yes</u> | 5. Candle wax burns. |
| <u>yes/no</u> | 6. Molten candle wax solidifies. |
| <u>yes</u> | 7. Green leaves turn yellow and red as the seasons change. |
| <u>yes</u> | 8. Baking powder produces a gas that makes a cake rise. |

In your textbook, read about how to represent chemical reactions and how to balance chemical equations.

Use the terms below to complete the passage. Each term may be used once, more than once, or not at all.

arrow	plus sign	(s)	(l)
reactant	product	(g)	(aq)

The fuel for the space shuttle is hydrogen, which burns in oxygen to produce water vapor and energy. In this chemical reaction, hydrogen is a(n) **(9)** reactant, oxygen is a(n) **(10)** reactant, and water vapor is a(n) **(11)** product. In a chemical equation for this reaction, a(n) **(12)** arrow is used to separate hydrogen and oxygen from water vapor and energy. A(n) **(13)** plus sign is used to separate the symbols for hydrogen and oxygen. A(n) **(14)** (g) symbol is used to tell the state of hydrogen in the reaction, a(n) **(15)** (g) symbol is used for the state of oxygen, and a(n) **(16)** (g) symbol is used for the state of water vapor.

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Seager/Slabaugh



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Study Guide and Solutions Manual for Fundamentals of General, Organic, and Biological Chemistry Susan McMurry, 1992

Study Guide and Solutions Manual, Fundamentals of General, Organic, and Biological Chemistry, Third Edition John McMurry, Susan McMurry, 1999 Provides worked out solutions to text problems along with chapter by chapter outlines and a variety of self tests at the end of each chapter

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 - The Rise of Digital Reading Chemical Reaction Study Guide Chemistry Answer
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemical Reaction Study Guide Chemistry Answer
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemical Reaction Study Guide Chemistry Answer
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemical Reaction Study Guide Chemistry Answer

- Personalized Recommendations
 - Chemical Reaction Study Guide Chemistry Answer User Reviews and Ratings
 - Chemical Reaction Study Guide Chemistry Answer and Bestseller Lists
5. Accessing Chemical Reaction Study Guide Chemistry Answer Free and Paid eBooks
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 - Chemical Reaction Study Guide Chemistry Answer eBook Subscription Services
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 6. Navigating Chemical Reaction Study Guide Chemistry Answer eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemical Reaction Study Guide Chemistry Answer Compatibility with Devices
 - Chemical Reaction Study Guide Chemistry Answer Enhanced eBook Features
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 - Adjustable Fonts and Text Sizes of Chemical Reaction Study Guide Chemistry Answer
 - Highlighting and Note-Taking Chemical Reaction Study Guide Chemistry Answer
 - Interactive Elements Chemical Reaction Study Guide Chemistry Answer
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 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemical Reaction Study Guide Chemistry Answer
 9. Balancing eBooks and Physical Books Chemical Reaction Study Guide Chemistry Answer
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemical Reaction Study Guide Chemistry Answer
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
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 - Setting Reading Goals Chemical Reaction Study Guide Chemistry Answer
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Chemical Reaction Study Guide Chemistry Answer

- Fact-Checking eBook Content of Chemical Reaction Study Guide Chemistry Answer
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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