

Chemistry Study Guide

Periodic Table of the Elements																		
H	Li	Be																
Na	Mg																	
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr	
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe	
Cs	Ba	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn		
Fr	Ra	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Uut	Fl	Uup	Lv	Uus	Uuo		
		La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu		
		Ac	Th	Pa	U	Ng	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr		

Objectives:

To understand the structure of an atom, and the nature + properties of subatomic particles.



Matter:

Anything that has mass and occupies space.

Examples: air, gases, liquids, solidz!

Atoms:

Atoms are the smallest, indivisible particle.

- Outer part → shell/orbit/energy level → carries the electronz (e-)
- Inner part → nucleus → houses the protonz (p+) and neutronz (n)
- Outer & Inner part → together these are the subatomic particles.

Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	p ⁺	1	+1	Nucleus
Neutron	n	1	0	Nucleus
Electron	e ⁻	-0	-1	Orbit/Shell

Chemistry 01 Study Guide

Karin Nielsen-Saines

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Table of Contents Chemistry 01 Study Guide

1. Understanding the eBook Chemistry 01 Study Guide
 - The Rise of Digital Reading Chemistry 01 Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry 01 Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry 01 Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry 01 Study Guide
 - Personalized Recommendations
 - Chemistry 01 Study Guide User Reviews and Ratings
 - Chemistry 01 Study Guide and Bestseller Lists
5. Accessing Chemistry 01 Study Guide Free and Paid eBooks

- Chemistry 01 Study Guide Public Domain eBooks
- Chemistry 01 Study Guide eBook Subscription Services
- Chemistry 01 Study Guide Budget-Friendly Options

6. Navigating Chemistry 01 Study Guide eBook Formats

- ePUB, PDF, MOBI, and More
- Chemistry 01 Study Guide Compatibility with Devices
- Chemistry 01 Study Guide Enhanced eBook Features

7. Enhancing Your Reading Experience

- Adjustable Fonts and Text Sizes of Chemistry 01 Study Guide
- Highlighting and Note-Taking Chemistry 01 Study Guide
- Interactive Elements Chemistry 01 Study Guide

8. Staying Engaged with Chemistry 01 Study Guide

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Chemistry 01 Study Guide

9. Balancing eBooks and Physical Books Chemistry 01 Study Guide

- Benefits of a Digital Library
- Creating a Diverse Reading Collection Chemistry 01 Study Guide

10. Overcoming Reading Challenges

- Dealing with Digital Eye Strain
- Minimizing Distractions
- Managing Screen Time

11. Cultivating a Reading Routine Chemistry 01 Study Guide

- Setting Reading Goals Chemistry 01 Study Guide
- Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Chemistry 01 Study Guide

- Fact-Checking eBook Content of Chemistry 01 Study Guide
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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