

# Challenging Calculations - Study-Guide

4. How many molecules are there in 2.48 moles of  $\text{CO}_2$ ?

It is 2.48 x  $10^{23}$  molecules

5. How many formula units are there in 2.5 moles of  $\text{CaBr}_2$ ?

It is 2.5 x  $10^{23}$  fu

6. How many molecules are there in 8 moles of hydrogen azide,  $\text{HN}_3$ ?

It is molecules

7. How many formula units of oxygen is?

It is 20 g grams

8. How many formula units of oxygen is?

It is 2.5 g grams

9. What is the mass formula units of  $\text{CaCl}_2$ ?

It is 2.5 g grams

10. What is the number of moles in 2.5 grams of  $\text{CaCl}_2$ ?

It is 2.5 moles of  $\text{CaCl}_2$

11. How many formula units of  $\text{CaCl}_2$  are there in 2.5 g. What is the number of moles of  $\text{CaCl}_2$ ?

It is 2.5 g grams

It is 2.5 g grams

12. If 2.5 g of  $\text{CaCl}_2$  is dissolved in 2.5 g of water, what is the mass of the solution?

It is 2.5 g grams

13. What is the mass of  $\text{CaCl}_2$  in 2.5 g of  $\text{CaCl}_2$ ?

It is 2.5 g grams

14. What is the mass of  $\text{CaCl}_2$  in 2.5 g of  $\text{CaCl}_2$ ?

It is 2.5 g grams

# Chemistry Chemical Quantities Study Guide Answers

**Burns A. Ralph, Charles Baker, Claire  
Baker**



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