

Study Guide for Chemistry EOC Assessment

Your test will consist of multiple-choice questions over the following topics.

1. Metric units for mass, length, volume
2. Metric to metric conversions (K H Da m/l/g D C M)
3. Know how to put numbers in scientific notation and take them out.
4. Significant Digit rules
5. Difference between chemical and physical properties and changes. There will be multiple questions on this.
6. Ways to tell if a chemical change (chemical reaction) is taking place.
7. Pure substances and mixtures
 - a. heterogeneous mixture
 - b. homogeneous mixture
8. Density & how it relates to floating and sinking
9. Use the equation for density
$$\text{Density} = \frac{\text{Mass}}{\text{Volume}}$$
$$(\text{g/ml}) \quad (\text{g}) \quad (\text{ml})$$
10. Know the phases of matter and the characteristics of each phase.
11. Law of Conservation of Matter
12. What is a mole in chemistry?
13. Convert grams to moles, and moles to grams
14. Define an atom
15. Structure of an atom and where each subatomic particle is located.
 - a. Nucleus (contains protons and neutrons)
 - b. Protons (positive)
 - c. Neutrons (neutral)
 - d. Electrons (negative)
16. Isotopes
17. Elements
18. If given atomic # and mass, find the number of protons and neutrons in an atom.
19. Find the number of valence electrons, and tell what it says about reactivity.
20. Periodic Table
 - a. Know about the families
 - b. Group vs. period
21. Know about the properties of metals and nonmetals.
 - a. Ductile, malleable, conductors
22. Ionic & Covalent Compounds
 - a. How do they form? (Transfer or gain electrons)
 - b. Characteristics of Each
23. Be able to name ionic compounds
24. Predict oxidation numbers
25. Chemical Equations:
 - a. Know the parts of an equation
 - b. 5 main types of reactions-know about them and be able to identify them (Synthesis, Decomposition, Single Displacement, Double Displacement, Combustion)
26. Acids & Bases:
 - a. Macro and Microscopic properties of each
 - b. What ions do they form, how do they affect litmus, what is their pH range?
 - c. Identify examples of each- Mono, di, and polyprotic acids
 - d. Tell if they are strong or weak
 - e. Acid-base indicator

Chemistry Eoc Study Guide Answers

Guillaume Favre

Chemistry Eoc Study Guide Answers:

Cracking the Virginia SOL Paul Foglino, Princeton Review (Firm), 2000 The Princeton Review realizes that acing the Chemistry exam is very different from getting straight As in school. The Princeton Review doesn't try to teach students everything there is to know about chemistry; only the techniques they'll need to score higher on the exam. There's a big difference. In **Cracking the Virginia SOL** EOC Chemistry, TPR will teach test takers how to think like the test makers and score higher by mastering what test takers need to know with proven techniques. Use word association to solve problems when unsure of the correct answers. Ace the exam by understanding the logic of the test makers and applying it to solutions. This book includes 2 full length simulated exams. All of TPR's sample test questions are just like the ones test takers will see on the actual End of Course Chemistry exam, and TPR fully explains every solution. Contents include: The Mystery Exams About the SOL Exams, How to Crack the Chemistry EOC, II Chemistry Review, Atomic Structure, Anatomy of the Periodic Table, Chemical Compounds, Chemical Reactions, Chemistry, Math, Gases, Phase Changes, Solutions, Equilibrium, Acids and Bases, How Chemical Reactions Happen, Laboratory III, The Princeton Review Practice Tests.

Study Guide and Solutions Manual

for Seager/Slabaugh's Chemistry for Today Seager/Slabaugh, 2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text. This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises. New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application.

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