

Answer Key

60 + 4 ext

Acids and Bases

How do acids and bases behave in water?

Why?

Acids and bases play an important role in our lives. Numerous biological processes, industrial applications, and even environmental problems are a function of the acidity or basicity (alkalinity) of aqueous solutions. It is therefore important to understand what makes a substance behave as an acid or a base when dissolved in water. In this activity, we will explore the physical and chemical properties of acids and bases.

Model 1 – Arrhenius Acids and Bases

Common Name for Aqueous Solution	Chemical Formula	Found in...	Tastes...	Turns Litmus Paper...	Conducts Electricity?	Acid or Base
Acetic acid	$\text{HC}_2\text{H}_3\text{O}_2(\text{aq})$	Vinegar	Sour	Red	Yes	Acid
Benzoic acid	$\text{HC}_7\text{H}_5\text{O}_2(\text{aq})$	Food preservative	Sour	Red	Yes	Acid
Phosphoric acid	$\text{H}_3\text{PO}_4(\text{aq})$	Soda pop	Sour	Red	Yes	Acid
Hydrochloric acid	$\text{HCl}(\text{aq})$	Stomach acid	Sour	Red	Yes	Acid
Citric acid	$\text{H}_3\text{C}_6\text{H}_7\text{O}_6(\text{aq})$	Citrus fruits	Sour	Red	Yes	Acid
Ascorbic acid	$\text{H}_2\text{C}_6\text{H}_8\text{O}_6(\text{aq})$	Vitamin C	Sour	Red	Yes	Acid
Magnesium hydroxide	$\text{Mg}(\text{OH})_2(\text{aq})$	Milk of magnesia	Bitter	Blue	Yes	Base
Aluminum hydroxide	$\text{Al}(\text{OH})_3(\text{aq})$	Antacids	Bitter	Blue	Yes	Base
Barium hydroxide	$\text{Ba}(\text{OH})_2(\text{aq})$	Lubricants	POISON	Blue	Yes	Base
Sodium hydroxide	$\text{NaOH}(\text{aq})$	Drain cleaner	POISON	Blue	Yes	Base

1. Refer to Model 1.
 - a. What is the common chemical name for Vitamin C?
Ascorbic acid
 - b. Is Vitamin C classified as an acid or a base?
Acid
2. Examine the properties of the Arrhenius acids in Model 1. List three properties that all Arrhenius acids have in common.
sour, turn litmus paper red, conduct electricity
3. Examine the chemical formulas for the Arrhenius acids in Model 1. What feature do all the Arrhenius acid chemical formulas have in common?
H at the beginning
4. Examine the properties of the Arrhenius bases in Model 1. List two properties that all Arrhenius bases have in common.
bitter / poison, turn litmus paper blue, conduct electricity

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**Donald Pritchard Smith, Henry Keppeler
Miller**

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explains the term of molecular sieves currently used to name these materials. Indeed a large number of industrial processes for separation of gases and liquids and for catalysis are based upon the use of this low dimensional feature in zeolites. For instance zeolites constitute the first class of catalysts employed all over the world. Because of the peculiarity and flexibility of their structure and composition zeolites can be adapted to suit many specific and diversified applications. For this reason zeolites are presently the object of a large and fast growing interest among chemists and chemical engineers.

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