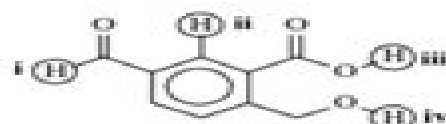


1. You have separate test tubes containing each of the following substances. You analyze each substance using broad-range pH paper. You compare the color of the paper after the analysis to the reference. Which of the following would you anticipate showing a pH of greater than 7?

aqueous baking soda	0.01 <i>N</i> HCl	ammonia-based cleaner	lemon juice	0.05 <i>N</i> NaOH
A	B	C	D	E

- A, C, and E would have a pH > 7
 - B and C would have a pH > 7
 - only C and E would have a pH > 7
 - all of the substances would have a pH > 7
 - none of the substances would have a pH > 7
2. In the Preparation of Aspirin, iron (III) chloride test positive (purple) for
- a carbonyl
 - an acetyl group
 - a phenol
 - an ester
 - an amide
3. By definition a melting point is:
- The temperature when solid-liquid equilibrium is reached.
 - The temperature when liquid-vapor equilibrium is reached.
 - The temperature when a solid decomposes.
 - The temperature at which sublimation occurs.
 - none of these
4. Casein is a _____ that is found in _____.
- protein / vegetable oil
 - enzyme / fruits
 - carbohydrate / milk
 - lipid / animal fat
 - protein / milk
5. If a substance gives an R_f value of 0.50 and the solvent front traveled 12.6 cm from the origin, then how far did the sample spot travel from the origin?
- 6.3 cm
 - 25.2 cm
 - 6.0 cm
 - 1.0 cm
 - none of these
6. Examine the following compound and determine which of the circled H's (*i.e.*, protons) you would expect to be the most acidic.



- i would be the most acidic
- ii would be the most acidic
- iii would be the most acidic
- iv would be the most acidic
- iii and iv are identical in acidity and are the most acidic

Chemistry Lab Final Exam Questions

Yicheng Fang



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