

# Unit One Chemistry Exam

## Multiple Choice 20 marks

You need to record your answers in the appropriate section in the ANSWER BOOKLET.

### Question 1

A nanoparticle is best described as

- A. one atom.
- B. one thousand atoms.
- C. one mole of atoms.
- D. all of the above are correct.

### Question 2

An element in group 2 and period 3 of the periodic table could have an electron configuration of

- A.  $1s^2 2s^2 2p^6 3s^2 3p^2$
- B.  $1s^2 2s^2 2p^6 3s^2$  or  $1s^2 2s^2 2p^6 3s^2 3p^2$
- C.  $3s^2$
- D.  $1s^2 2s^2 2p^1$

### Question 3

An emission spectrum is produced when the valence electrons of an atom

- A. become excited.
- B. move back down to a lower energy level.
- C. remain in the ground state.
- D. undergo a transition that brings them further away from the nucleus.

### Question 4

Avogadro's number is best described as

- A. the number of atoms in one mole of carbon-12.
- B. the number of atoms present in 12.011 g of the carbon – 12 isotope.
- C. the number of grams of the carbon – 12 isotope present in one mole of carbon.
- D. the number of atoms present in 12 g of any element.

### Question 5

The relative molecular mass of a compound is assigned the units

- A.  $\text{kJ mol}^{-1}$
- B.  $\text{g mol}^{-1}$
- C. mol
- D. none of the above

## Chemistry Question Form 6

**Camilla Rothe**

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