

# Chemistry Spring Final Study Guide

### Unit 10: Acids & Bases

Use the following information and complete the structure of Acids & Bases.

Acids	Bases
<p><b>Unit 10: Acids &amp; Bases</b></p> <p>Draw a line on the graph and indicate its dependence for each property: Temperature, Volume, and Pressure below.</p> <p>Label the y-axis.</p> <p>Label the x-axis.</p> <p>1. A student of the periodic table.</p> <p>2. A student of the periodic table.</p> <p>3. A student of the periodic table.</p> <p>4. The gas in air.</p> <p>5. The common.</p>	<p><b>Unit 10: Acids &amp; Bases</b></p> <p>Draw the following:</p> <p>1. Draw the structure of the following:</p> <p>2. Draw the structure of the following:</p> <p>3. Draw the structure of the following:</p> <p>4. Draw the structure of the following:</p> <p>5. Draw the structure of the following:</p>

### Chemistry Spring Final Study Guide

Name: \_\_\_\_\_

**Unit 1: Measurement & Analysis**

Mark the best answer from those provided to each question.

Mark the best answer from those provided to each question.

Determine the number of Significant Figures in each of the following numbers:

Mass = 1.23456789 g	Mass = 1.23456789 g
Volume = 1.23456789 L	Volume = 1.23456789 L

Convert the following into Scientific Notation:

Mass = 1.23456789 g	Mass = 1.23456789 g
Volume = 1.23456789 L	Volume = 1.23456789 L

**Unit 2: Matter & Energy**

Mark the best answer from those provided to each question.

Mark the best answer from those provided to each question.

Determine the number of Significant Figures in each of the following numbers:

Mass = 1.23456789 g	Mass = 1.23456789 g
Volume = 1.23456789 L	Volume = 1.23456789 L

Convert the following into Scientific Notation:

Mass = 1.23456789 g	Mass = 1.23456789 g
Volume = 1.23456789 L	Volume = 1.23456789 L

**Unit 3: Matter & Energy**

Mark the best answer from those provided to each question.

Mark the best answer from those provided to each question.

Determine the number of Significant Figures in each of the following numbers:

Mass = 1.23456789 g	Mass = 1.23456789 g
Volume = 1.23456789 L	Volume = 1.23456789 L

Convert the following into Scientific Notation:

Mass = 1.23456789 g	Mass = 1.23456789 g
Volume = 1.23456789 L	Volume = 1.23456789 L

**Unit 4: Matter & Energy**

Mark the best answer from those provided to each question.

Mark the best answer from those provided to each question.

Determine the number of Significant Figures in each of the following numbers:

Mass = 1.23456789 g	Mass = 1.23456789 g
Volume = 1.23456789 L	Volume = 1.23456789 L

Convert the following into Scientific Notation:

Mass = 1.23456789 g	Mass = 1.23456789 g
Volume = 1.23456789 L	Volume = 1.23456789 L

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**LP Steffe**



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