

Study Guide Unit 9

1. KNOW the following vocabulary words (this means you NEED to be able to DEFINE them ALL):

- a. Solution-
m. Solute-
- b. Homogeneous-
n. Solvent-
- c. Heterogeneous-
o. Acid-
- d. Soluble-
p. Base-
- e. Insoluble-
q. Titration-
- f. Miscible-
r. Analyte-
- g. Immiscible-
s. Titrant-
- h. Solubility-
t. Monoprotic acid-
- i. Alloy-
u. Diprotic Acid
- j. Saturated-
v. Triprotic Acid
- k. Unsaturated-
- l. Supersaturated-

2. How is the solubility of a gas affected by temp. and pressure?

Pressure up = solubility up

Temp up = solubility down

3. Under what conditions are salts most soluble?

High temperatures

4. Use the solubility curve to the right to answer the following.

a. What mass of KNO_3 will make a saturated solution at 60°C ? **100g**

b. Which of the compounds shown on the solubility curve decrease solubility as temperature increases?

NH_3 , $\text{Ca}_3(\text{SO}_4)_2$

c. What type of solution do I have if I **add** 110 g of NaNO_3 in 100g of 20°C water?

Saturated

d. What mass of KCl can go into solution at 80°C ?

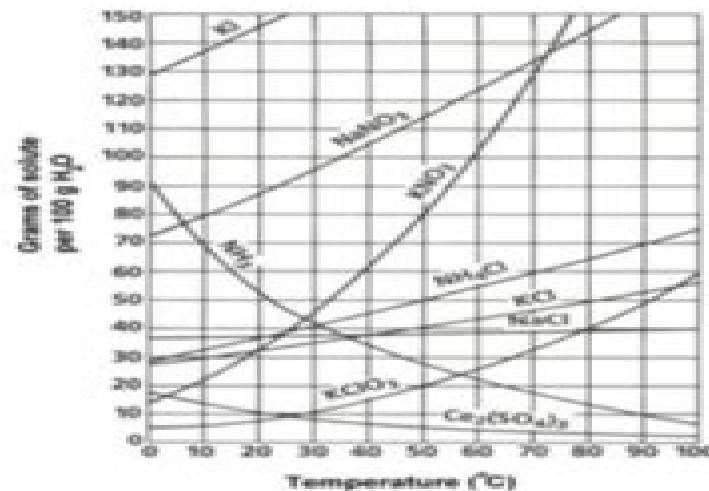
50 g

e. What is the least soluble compound at 20°C ?

KClO_3

f. Excluding KI , what is the most soluble compound at 80°C ?

KNO_3



Chemistry Study Guide Answers What Are Solutions

Darrell D. Ebbing

Chemistry Study Guide Answers What Are Solutions:

Organic Chemistry Study Guide and Solutions Manual Marc Loudon, Jim Parise, Lisa Bonner, 2024-08-09 The Study Guide and Solutions Manual offers several study tools for students. Study Guide Links are brief supplements to text material and provide hints for study shortcuts approaches to problem solving or additional explanations of topics that typically cause difficulty. Further Explorations are short discussions that provide additional depth on content in the text. Reaction Reviews present the reactions in the chapter in detailed outline form. This summary consolidates the reactions of the chapter in one place and presents not only the reactions themselves but also the essential features of their mechanisms and stereochemistry for cases in which these issues are discussed in the text. Solutions are provided for all of the over 1600 problems in the text. Many include worked out examples in which not only the answer is given but also the detailed approach to the solution is discussed.

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Study Guide and Solutions Manual for McMurry's Organic Chemistry, Fifth Edition Susan McMurry, 2000

Provides answers and explanations to all in text and end of chapter exercises. Also includes summaries of name reactions, functional group synthesis and reactions, lists of reagents and abbreviations, and articles on topics ranging from infrared absorption frequencies to the Nobel Prize winners in Chemistry. This edition now includes all new artwork expanded in text problems, summary quizzes approximately every three chapters, more detailed explanations in solutions and chapter outlines.

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Student Study Guide and Solutions Manual to accompany General Organic and Biological Chemistry, 1e Kenneth W. Raymond, 2005-10-07

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