

# Chemistry Spring Final Study Guide

**Unit 10: Acids & Bases**

Find the pH of the following solutions and determine the concentration of acids & bases.

1. Find the pH of 0.1 M HCl.

2. Find the pH of 0.1 M NaOH.

3. Find the pH of 0.1 M  $\text{CH}_3\text{COOH}$ .

4. Find the pH of 0.1 M  $\text{NH}_4\text{Cl}$ .

5. Find the pH of 0.1 M  $\text{H}_2\text{SO}_4$ .

6. Find the pH of 0.1 M  $\text{Ba(OH)}_2$ .

7. Find the pH of 0.1 M  $\text{HNO}_3$ .

8. Find the pH of 0.1 M  $\text{KOH}$ .

9. Find the pH of 0.1 M  $\text{H}_2\text{SO}_4$ .

10. Find the pH of 0.1 M  $\text{Ba(OH)}_2$ .

**Unit 11: Chemical Equilibrium**

Write the chemical formula for each compound.

1. Sodium chloride

2. Potassium nitrate

3. Calcium hydroxide

4. Ammonium sulfate

5. Magnesium oxide

6. Zinc sulfide

7. Barium carbonate

8. Aluminum hydroxide

9. Copper(II) sulfate

10. Silver nitrate

11. Lead(II) acetate

12. Nickel(II) chloride

13. Cobalt(II) bromide

14. Manganese(II) iodide

15. Chromium(III) phosphate

16. Vanadium(V) oxide

17. Tin(IV) selenide

18. Bismuth(III) nitrate

19. Antimony(III) chloride

20. Arsenic(III) bromide

21. Tellurium(IV) selenide

22. Molybdenum(V) nitrate

23. Niobium(V) pentachloride

24. Vanadium(V) pentachloride

25. Antimony(V) pentachloride

26. Bismuth(V) pentachloride

27. Tellurium(V) pentachloride

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**L Reisser**



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