

Conclusions/Post Lab Questions

The experiment was performed in order to recrystallize Benzoic acid from a mixture containing Si_2O (Sand), Benzoic Acid, and NaCl . An iota was taken from the original mixture which weighed 0.5106 g. After calculating the percentage of Benzoic acid which was 32.8% of the mixture taken. Then it was used in order to find the initial mass which was calculated to be 0.1675g of the 0.5106g used. The mixture then had acetone added to it so that the Benzoic acid which is very soluble in acetone and NaCl which is semi soluble could be dissolved into it while the Si_2O (Sand) could not because it was insoluble in acetone which then separated that from the mixture. Next is to gravity filter out the acetone which contains the dissolved Benzoic acid and some of the NaCl . During this process of gravity filtering some of the acetone got onto the rim of the funnel and the outside of the rim which resulted in possible loss of Benzoic acid. The acetone is then boiled off and the NaCl and Benzoic acid remain while after it is all boiled off DI water is added to redissolve the NaCl and Benzoic acid. After it has fully dissolved it was then placed on the counter to air cool at room temperature which helped the Benzoic acid recrystallize without any impurities inside of it. To induce further recrystallization and prevent the Benzoic acid from redissolving into the DI water it is placed into an ice bath to stay cold. Vacuum filtration was then performed with a buncher flask after wetting the filter with DI water to make sure none of the required material was lost. As a result, 0.0865g of Benzoic acid were recovered which was a recovery percentage of 51.6%. The amount recovered was less than what was actually recovered due to a mistake made when looking at the order of the lab and doing the melting point before measuring the weight of the recovered material. Another possible loss was when packing the glass tubes for the melting point a bit of the recovered material got onto the table and was unable to be recovered. A melting point test was then used and tube 1 has a start point of 120 degrees Celsius and an end point of 124 degrees Celsius which was calculated to have an average of 122 degrees Celsius, and using the melting point obtained from PubChem an error rate of below 0% was calculated. Tube 2 had a starting point of 121 degrees Celsius and an end point of 124 degrees Celsius which averaged out to a melting point of 122.5 degrees Celsius which calculated to have an error percentage of also below 0% using PubChem's melting point of 122.4 degrees Celsius. Such results showed the Benzoic acid to be very pure even though only 51.3% was recovered.

Possibilities for the diminished amount recovered could be due to the spilled mixture when it was being grinded and mixed, the left-over acetone on the funnel and the outside of the funnel, when it was weighed after the melting point was taken so some of the product was not able to be weighed, poor rinsing of glassware before moving to the next part, and the amount that was spilled onto the table. Ways of reducing such losses would be to more carefully and skillfully transfer the acetone from one baker to another. Another way of reducing lost material is to read and follow the directions better as to not forget to weigh the recovered material before doing the melting point part of the experiment.

Conclusion For Organic Models Lab

G Orfield



Conclusion For Organic Models Lab:

Laboratory Manual of Pharmaceutical Organic Chemistry I Mr. Sushant Sudhir Pande, Dr. Prashant D. Aragade, Dr. Saurabh C. Khadse, Dr. Niranjana S. Mahajan, 2024-02-16 We are pleased to put forth the Laboratory Manual of Pharmaceutical Organic Chemistry I This manual prepared according to the PCI B Pharm course regulations 2014 is divided into three sections systematic qualitative analysis preparation of suitable solid derivatives and construction of molecular models The methods of all the experiments are drawn from the latest editions of official books of pharmaceutical organic chemistry and research papers ensuring the inclusion of the latest advancements in methodologies or apparatus This manual is designed for outcome based education Each experiment follows a uniform format with sections for practical significance practical outcomes PrOs mapping with course outcomes theory resources used procedure precautions observations results conclusion references and synopsis questions Each experiment offers an opportunity for students to perform practical work developing proficiency in effectively managing equipment handling glassware chemicals reagents and writing analytical reports In addition the questions at the end of the experiments help to enhance students knowledge benefiting them as they pursue higher studies During the laboratory period you will have to multiple tasks while performing the experiment It is essential to document your actions and observations thoroughly as you proceed Always plan your work ahead considering what you are doing why you are doing it what is happening and what conclusions you can draw from your experiment We acknowledge the help and cooperation of various individuals in bringing out this manual We are highly indebted to the authors of the books and articles mentioned in the references which were a major source of information for this manual We also thank the publishers designers and printers who worked hard to publish this manual in a timely manner We hope that this manual will be helpful to students in understanding concepts principles and performing procedures We wish you all the best *Introduction to Organic Research* Ebenezer Emmet Reid, 1924 **Transport and Fate of Acetone in an Outdoor Model Stream, Stennis Space Center Near Bay St. Louis, Mississippi**, 1990 Studies on Life at the Energetic Edge - from Laboratory Experiments to Field-Based Investigations, Volume II Mark Alexander Lever, Bo Barker Jorgensen, Jan Amend, Victoria Orphan, Tori Hoehler, 2024-01-16 In collaboration with Microenergy 2022 The 4th International Workshop on Microbial Life under Extreme Energy Limitation we are proud to launch Volume II of Studies on Life at the Energetic Edge from Laboratory Experiments to Field Based Investigations This workshop focuses on the energy controls on microbial life and the exploration of the biological demand for energy Genetic adaptations and phenotypic traits that enable microorganisms to tolerate long periods of energy limitation have attracted broad scientific interest in recent years Laboratory based cultivation experiments have shown that the potential to survive weeks to months in the absence of energy inputs occurs across a phylogenetically wide range of microbes Studies on natural environments have shown that energy limitation is pervasive across most habitats on Earth from highly metabolically active surface habitats to subsurface

environments that have been cut off from new energy inputs for thousands of years Yet much remains to be learned about the evolutionary adaptations and life history traits that enable microorganisms to live under low energy conditions Similarly the spectrum of energy sources and metabolisms that enable and support life on Earth and potentially elsewhere in the Universe is far from constrained *Introduction to Organic Laboratory Techniques* Donald L. Pavia, 2007 Resumen Taking an organic chemistry laboratory course You need a manual you can trust This proven laboratory manual gives you what you need to conduct a variety of interesting microscale experiments with safety and ease while you develop an understanding of the special techniques these type of experiments require The authors have increased the book's green approach giving you the clearly written information and instruction to conduct chemical experiments in a more environmentally friendly way Many of the book's experiments have been modified to use new techniques and reduce the use of hazardous solvents and reagents You'll find fascinating essays that add real life relevance and understanding to each experiment including Identification of Drugs Petroleum and Fossil Fuels Detection of Alcohol The Breathalyzer and Fireflies and Photochemistry

Contribution from the Research Laboratory of Organic Chemistry Massachusetts Institute of Technology. Research Laboratory of Organic Chemistry, 1929 **Report summaries** United States. Environmental Protection Agency, 1983 **A Laboratory Manual of Organic Chemistry for Medical Students** Matthew Steel, 1920 Multiphase Flow and Transport Models for Organic Chemicals, 1988 *Construction, Implementation, and Evaluation of an Undergraduate Biology Laboratory Teaching Model* Todd M. Tarrant, 2005 **Laboratory Planetology** COSPAR. Scientific Assembly, 1995 *Environmental Toxicology and Chemistry*, 2007 Sapira's Art and Science of Bedside Diagnosis Jane M.

Orient, 2012-03-28 The Fourth Edition of this textbook teaches the artful science of the patient interview and the physical examination Chapters are filled with clinical pearls vignettes step by step methods and explanations of the physiologic significance of findings New features include Points to Remember over 300 questions with answers and discussion over 120 additional references and expanded discussions of the usage and pitfalls of evidence based medicine Other highlights include expanded and updated discussions of sleep apnea minor head trauma cervical spine involvement in rheumatoid arthritis transplantation related problems adverse effects of AIDS therapy and more A companion Website includes fully searchable text and a 300 question test bank Laboratory Manual for a Short Course in Organic Chemistry Ernest Edwin

Campaigne, Robert D. Schuetz, 1953 **Publications of the Jet Propulsion Laboratory**, 1964 *Organic Chemistry Education Research into Practice* Jay Wackerly, Sarah Zingales, Michael Wentzel, Gautam Bhattacharyya, Brett

McCollum, 2025-03-25 This Research Topic has three main goals 1 provide a platform for instructors of organic chemistry to showcase evidence based methods and educational theories they have utilized in their classrooms 2 build new and strengthen existing connections between educational researchers and practitioners and 3 highlight how people have used chemical education based research in their teaching practice There are places in the literature dedicated for chemical

education research CER however there is not a clear avenue for those that have changed their teaching methods based on published CER and report their experiences Creating this article collection will foster collaboration between chemical education researchers and teachers of organic chemistry This opportunity allows these instructors to share evidence based practices experiences challenges and innovative approaches from CER literature and beyond This Research Topic bridges discipline based education research and the scholarship of teaching and learning which will help advance organic chemistry education and improve student outcomes *Laboratory Studies on the Formation of Secondary Organic Aerosol from the Atmospheric Oxidation of Alkenes* Kenneth Stephen Docherty,2004 Meteorological and Geostrophysical Abstracts ,1986
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