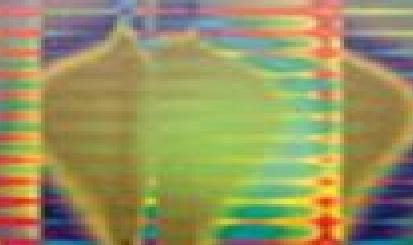


# Computer Structure and Logic



# Computer Structure And Logic Study Guide

**Jiyuan Zhang**

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web 1 approximately 3 g of hydrated oxalic acid  $\text{H}_2\text{C}_2\text{O}_4 \cdot 2\text{H}_2\text{O}$  was weighted to the nearest 0 g in a 50 ml beaker 2 about 30 ml of distilled water was added to the beaker to dissolve the oxalic acid 3 the solution was transferred into 250 ml volumetric flask the beaker was rinsed and its contents were poured into the flask

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web solution a titrant volume 0 ml the solution ph is due to the acid ionization of  $\text{HCl}$  because this is a strong acid the ionization is complete and the hydronium ion molarity is 0 100 m the ph of the solution is then  $\text{pH} = \log 0.100 = 1.000$   $\text{pH} = \log 0.100 = 1.000$  b titrant volume 12 50 ml

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